Sensitive Assay for Carvedilol in Tablets and Spiked Human Plasma Using a Flow-Injection Chemiluminescent Method

Nawal Al Arfaj1, Heba H. Abdine2, Maha A. Sultan2

1Department of Chemistry, College of Science, Women Student-Medical Studies and Sciences Sections, King Saud University, P.O. Box 22452, Riyadh 11495, Saudi Arabia; 2Department of Pharmaceutical Chemistry, College of Pharmacy, Women Student-Medical Studies and Sciences Sections, King Saud University, P.O. Box 22452, Riyadh 11495, Saudi Arabia

ABSTRACT

A simple and sensitive chemiluminescent method using flow injection (FI) is developed for the determination of carvedilol, based on the reaction of carvedilol with tris (2, 2'-bipyridyl) ruthenium (II), and KMnO4 in sulfuric acid medium. Under the optimum conditions; the chemiluminescence (CL) intensity is a linear function of carvedilol concentration over the range of 0.04-1.0 μg ml⁻¹ (9.8 x 10⁻⁸ - 2.5 x 10⁻⁶ mol L⁻¹) with a detection limit (S/N=3) of 0.025 μg ml⁻¹ (6.2 x 10⁻⁸ mol L⁻¹). The relative standard deviation of the proposed method calculated from 10 replicate injections of 0.4 μg ml⁻¹ carvedilol is 0.95%. The sample throughput is 90 samples h⁻¹. The method is applied successfully to the determination of carvedilol in tablets dosage form and spiked human plasma.

Keywords: carvedilol; chemiluminescence; flow injection; Tris (2, 2'-bipyridyl) ruthenium (II); KMnO₄; tablets; human plasma

INTRODUCTION

Carvedilol (±)-1-(carbazol-4-yloxy)-3-[(2-(O-methoxy phenoxy)ethyl)amino]-2-propanol is a non-cardioselective β-blocker (1) (Fig. 1). It also has vasodilating properties. It is used in the management of hypertension and angina pectoris, and as an adjunct to standard therapy in symptomatic heart failure (1, 2).

The European Pharmacopoeia (3) describes a non-aqueous titration procedure for the determination of carvedilol in bulk.

Several methods are described for the determination of carvedilol in biological fluids including high performance liquid chromatography (4-9) and capillary electrophoresis (10-12). Few methods are reported for the analysis of carvedilol in tablets including fluorimetry (13, 14) and differential pulse voltammetry (15).

Although most of these methods are specific, their applications are limited due to the high commercial price. Chemiluminescence (CL) has received much attention as an attractive method for analytical application because of its high sensitivity, high selectivity, and small amount of chemical consumption, cost effectiveness, simple sample preparation and instrumentation (16). Only one procedure is reported in the literature for the determination of carvedilol in pharmaceutical dosage forms by using flow injection.

Figure 1. Structure of carvedilol.
CL ASSAY OF CARVEDILOL

The method is based on the chemiluminescence reaction of the drug with \([\text{Ru(bipy)}]^{2+}\) and potassium permanganate in sulfuric acid medium. Although, carvedilol has been determined using a multi-pumping flow system with the detection of the CL generated from the luminol system in presence of hypochlorite (17), the proposed method is easier to implement for routine analysis in pharmaceutical control using a simpler equipment. This straightforward method has shown its usefulness in the analysis of carvedilol in tablets and spiked human plasma.

EXPERIMENTAL

Apparatus and manifold

The flow system used for the determination and CL detection of carvedilol is shown schematically in Fig. 2. A Gilson minipuls 3M P4 peristaltic pump (two channels, variable speed) was used to drive the carrier and the reagent streams through the flow system. Each stream was pumped at a constant flow rate using PTFE tubing (0.8 mm i.d.).

The drug solution (100 \(\mu\)l) was injected through the sample injection valve, which allows the mixing of the sample with an acidified 3 \(\times\) 10\(^{-4}\) M \(\text{KMnO}_4\) solution and then combined with 1 \(\times\) 10\(^{-3}\) M \([\text{Ru(bipy)}]^{2+}\) solution just before the detector. The emitted light intensity was measured by a photomultiplier tube (PMT, THORN EMI 9789QB) which was operated at 1100 V. The PMT voltage was provided by a stable power supply (THORN EMI, Model PM 288BN).

The signal was recorded by a Yokogawa Model 3021 Recorder (Yokogawa, Japan). Peak height was measured for each signal and expressed as a voltage output of the photomultiplier tube.

Reagents

All the reagents used were of analytical reagent grade and the solutions were prepared with doubly distilled water. The following reagents were used: aqueous potassium permanganate (Fluka, UK), solution of 3 \(\times\) 10\(^{-4}\) M was prepared in 0.8 M sulfuric acid (Reidel-deHaen, Germany); aqueous \([\text{Ru(bipy)}]^{2+}\) (Aldrich, Chem. Co.) solution of 1 \(\times\) 10\(^{-3}\) M was prepared by dissolving \([\text{Ru(bipy)}]^{2+}\) hexahydrate in distilled water.

Materials

Pure drug samples, (99.8\%), were kindly supplied by Saudi Pharmaceutical Industries & Medical Appliances Corporation, Al-Qassim Pharmaceutical Plant, and Saudi Arabia (Dilatrend tablets (B.N.M 1003), each labeled to contain 25 mg carvedilol and manufactured by Saudi Pharmaceutical Industries under License of F. Hoffmann – La Roche Ltd, Basel, Switzerland by Roche S.P.A. Milan, Italy, were obtained from commercial sources.

Human plasma was obtained from King Khalid University Hospital, Blood Bank, King Saud University, and Riyadh and kept frozen until use after gentle thawing.

Preparation of standard solution

Stock solution (1.0 mg ml\(^{-1}\)) of carvedilol was prepared by dissolving 50.0 mg of carvedilol in methanol in a 50-ml measuring flask. The solution was protected from light and stored in a refrigerator at approximately 4 °C, it remained stable for at least 2 weeks. Working standard solutions with carvedilol concentrations ranging from 0.04 to 1.0 \(\mu\)g ml\(^{-1}\) were daily prepared by methanol dilutions of the above stock solution.

Recommended procedure for calibration

The FI manifold described in Fig. 2 was used. A 100 \(\mu\)l of drug solution was injected into a stream of acidified 3 \(\times\) 10\(^{-4}\) M \(\text{KMnO}_4\) solution which was then combined with a stream of 1 \(\times\) 10\(^{-3}\) M \([\text{Ru(bipy)}]^{2+}\) solution. The resulting peak height in mV was measured and plotted against drug concentration to obtain a calibration curve. Alternatively, the regression equation was derived.

Procedure for pharmaceutical formulations

Ten tablets were weighed and finally powdered. A weighed amount of the fine powder equivalent to 10 mg of carvedilol was dissolved in methanol by sonication for 10 min. The solution was filtered into a 100-ml volumetric flask and the filtrate was diluted to volume with methanol. This solution, labeled to contain 100 \(\mu\)g ml\(^{-1}\) of drug, was analyzed by the FI-CL procedure described above for calibration, and a standard addition method was used. Nominal content of tablets was calculated either from a previ-
CL ASSAY OF CARVEDILOL

ously plotted calibration graph or by using the regression equation.

Procedure for spiked human plasma

Different concentrations of standard carvedilol (0.1, 0.4 and 0.8 µg ml⁻¹) were added to 1 ml drug free plasma in centrifuge tube. 5 ml of methanol was added and vortexed for 2 min. The tube was centrifuged for 5 min at 2600 rpm. The supernatant was transferred to another tube and evaporated under nitrogen at 50 °C. 1 ml of methanol was added to the dry residue and the sample was dissolved by shaking for 30 s at 1500 rpm. Finally, the solution was centrifuged for 5 min at 2600 rpm and the supernatant was transferred to the polypropylene autosampler vial and was analyzed by the FI-CL procedure as described above for calibration. Nominal content of each plasma sample was calculated by using the regression equation for the drug in plasma previously calculated.

RESULTS AND DISCUSSION

Optimization of the flow injection system

Manifold design. A two-line manifold was used for the determination of carvedilol (Fig. 2). Maximum CL intensity was obtained when the sample was injected into a stream of acidified KMnO₄ and then mixed with Ru(bipy)₃²⁺ just before the detector.

Effect of different oxidants. Several oxidants such as K₂Cr₂O₇, Ce (IV), H₂O₂, Fe (III) and KMnO₄ were evaluated. It was found that KMnO₄ produced the highest CL intensity.

Effect of different concentrations of acids. HCl, HNO₃, H₃PO₄ and H₂SO₄ were used in different concentrations as diluent for KMnO₄ solution to study their effect on the CL intensity. The results showed that the maximum CL signal was attained with H₂SO₄. The influence of different concentrations of H₂SO₄ from 0.1–1.5 M was studied. It was found that the CL intensity increased with increasing H₂SO₄ concentration and reached maximum value at 0.8 M, (Fig. 3). Thus, a concentration of 0.80 M was selected for this work.

Effect of KMnO₄ concentration. The results of experiments showed that KMnO₄ concentration dramatically influenced the CL signals. So the concentration of KMnO₄ was studied ranging from 1 × 10⁻⁴ – 5 × 10⁻³ M. The peak height increased with raising KMnO₄ concentration up to 3 × 10⁻⁴ M above which the CL intensity decreased sharply. Probably, because of self absorption of the KMnO₄. Therefore, 3 × 10⁻⁴ M KMnO₄ was used for subsequent work (Fig. 4).

Effect of Ru(bipy)₃²⁺ concentration. The effect of the concentration of Ru(bipy)₃²⁺ was examined in the range of 1 × 10⁻⁴ – 5 × 10⁻³ M. The results showed that the CL intensity increased with an increase of Ru(bipy)₃²⁺ concentration. A 1 × 10⁻³ M was chosen as the best concentration that gave the highest CL intensity (Fig. 5).

Effect of Reagent flow rate. The total flow rates of Ru(bipy)₃²⁺ and acidified KMnO₄ streams were varied over the range 0.1 to 2.0 ml min⁻¹ with equal flows in each channel. The best total flow rate was found to be 0.7 ml min⁻¹ (0.35 ml min⁻¹ for each channel) as shown in Fig. 6.
Method validation

Under the optimized experimental conditions, the calibration curve was linear from 0.04-1.0 μg ml⁻¹. The regression equation was

\[ I = 67.763C + 0.5185 \quad (r = 0.9992, n = 7) \]

where \( I \) = intensity of the peak height (mV), \( C \) = concentration of carvedilol in μg ml⁻¹.

Statistical evaluation of the regression parameters gave the following values: standard deviation of the slope (Sₚ) 1.31, standard deviation of the intercept (Sᵦ) 0.42 and the residual standard deviation (Sₑᵦ) 0.64. These small figures point out to the high precision of the method (20). The perfect linearity of the calibration curve is clear by the correlation coefficient. Limit of detection (LOD) and limit of quantitation (LOQ) were estimated from the standard deviation of the blank. A LOD (S/N=3) of 0.025 μg ml⁻¹ and a LOQ of 0.035 μg ml⁻¹ were obtained. The linear range of the proposed method was similar to the reported chemiluminometric method (17), but the proposed method is direct and requires simpler equipment using one pump.

The accuracy of the method was tested with several synthetic samples of carvedilol with different concentrations over the range of 0.04-1.0 μg ml⁻¹ (n=7). The recovery is excellent and the results were in accordance with those given by the fluorimetric reference method (measuring the native fluorescence intensity at 356 nm with excitation at 254 nm) (14) as revealed by statistical analysis adopting Student’s t-test and F-test, where no significance difference was noticed between the two methods (Table 2).

Inter and intraday precision were assessed by analyzing three plasma samples containing different concentrations of the drug (Table 3) on the same day and for six days, respectively. The relative standard deviations were less than 5.9% and 7.2%, respectively.

Specificity of the proposed method for the determination of 0.4 μg ml⁻¹ carvedilol was examined in the presence of excipients (sucrose, lactose, povidone, magnesium stearate) in the same ratio of those used in tablet dosage form. These excipients had no effect on carvedilol CL response.

Application

The proposed CL method was applied to determine carvedilol in tablets. Table 2 shows that the results obtained are in good agreement with the claimed value and with those given by the reference method (14) as revealed by statistical analysis adopting Student’s t-test and F-test where no significant difference was noticed between the two methods.
A standard addition procedure was carried out in order to test the recovery of the proposed method. Thus, carvedilol concentration of 0.5 µg ml⁻¹ was added to known samples of tablets. After measuring the CL intensity, the recovery of each spiked sample was calculated (Table 4).

The high sensitivity of the proposed method allowed the method to be applied to the in-vitro determination of the drug in spiked human plasma. A prior deproteination step was necessary before application of the method. A calibration graph was first obtained by spiking plasma sample with carvedilol in the range 0.05–1.0 µg ml⁻¹. Linear regression analysis of the data gave this equation:

\[
\text{Carvedilol (µg ml}^{-1}\text{)} = 0.494 \pm 0.032
\]

Table 3. Inter and Intra-assay precision and accuracy for carvedilol in spiked human plasma

<table>
<thead>
<tr>
<th>Sample No.</th>
<th>Sample content (µg ml⁻¹)</th>
<th>Carvedilol (µg ml⁻¹)</th>
<th>Recovery * %</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>0.1</td>
<td>0.5</td>
<td>98.82</td>
</tr>
<tr>
<td>2</td>
<td>0.2</td>
<td>0.5</td>
<td>95.90</td>
</tr>
<tr>
<td>3</td>
<td>0.3</td>
<td>0.5</td>
<td>101.77</td>
</tr>
<tr>
<td>4</td>
<td>0.4</td>
<td>0.5</td>
<td>101.77</td>
</tr>
</tbody>
</table>

Mean (±S.D.) = 99.57 ± 2.43

* Each value is the mean of five determinations.
I = 54.325C + 1.0384 \quad (r = 0.9985)

where I is intensity of the peak height (mV) and C is the concentration of carvedilol in plasma in μg mL⁻¹. The lower limit of detection was found to be 0.04 μg mL⁻¹. The results of analysis of 3 spiked samples of plasma are presented in Table 3.

**Interference**

Ampicillin, which is often co-administered with carvedilol to treat heart trouble in the clinic (13), and hydrochlorothiazide, which is co-formulated with carvedilol, did not interfere with the assay of carvedilol as they have no CL response under the proposed conditions.

**Suggested CL mechanism**

Ru(bipy)²⁺ CL has proven to be a very sensitive detection system for compounds which contain amine (18). Thus the proposed reaction mechanism is thought to involve the oxidation of Ru(bipy)³⁺ and the N-heterocyclic nitrogen of the carbazole moiety and or the nitrogen of secondary amino group in the side chain of carvedilol by KMnO₄. The oxidation product of amine undergoes deprotonation to form a radical. This reduces the Ru(bipy)³⁺ to the excited state that subsequently emits light.

Carvedilol + KMnO₄ → Carvedilol²⁺ + Mn²⁺ + 4H₂O

Carvedilol²⁺ → Carvedilol¹⁺ + H⁺

Ru(bipy)³⁺ + Carvedilol¹⁺ + H₂O → [Ru(bipy)²⁺]¹⁺ + Carvedilol¹⁺ + H⁺

[Ru(bipy)²⁺]¹⁺ → Ru(bipy)²⁺ + Light

**CONCLUSION**

A flow-injection CL method has been established for the determination of carvedilol based on the reaction of the drug with Ru(bipy)³⁺ and KMnO₄ in sulfuric acid medium. The method has the merits of high sensitivity, selectivity, precision, wide linear range and a short analytical time. Consequently, the method can be used for carvedilol determination in pharmaceuticals and human plasma in presence of ampicillin without prior separation, as there is no interference from ampicillin on the determination of carvedilol by the proposed CL-method. The developed method offers a similar linear range comparatively to the reported chemiluminometric method (17), but the proposed method is simpler; using a single pump and it could be used for the determination of carvedilol in presence of ampicillin and in human plasma.

**REFERENCES**