

CLS 291 Clinical Hematology 1

جامعة  
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**Lecture 2**

# **Haemoglobinometry**

# Outlines

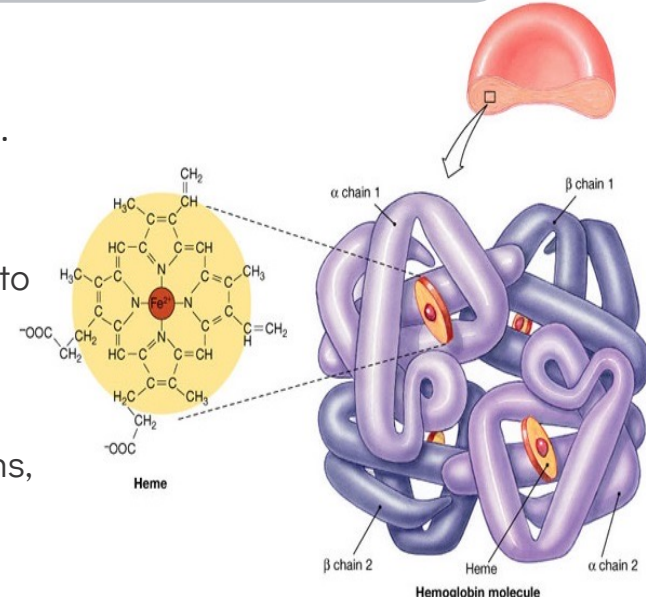
- I. Difference between Manual and Automated procedures.
- II. Hemoglobin protein overview.
- III. Haemoglobinometry: haemiglobincyanide.
- IV. Spectrophotometry.
- V. Using the Micropipette.
- VI. The purpose of using a blank
- VII. Result calculation and interpretation.

# Manual VS Automated Analysis



# The Hemoglobin Protein

- Hemoglobin (Hb) = globin (protein) + haemochromogen (Fe (II)) complex).
- Function:
  - The primary function of Hb is to transport oxygen (O<sub>2</sub>) from the lung to tissues, binding and releasing O<sub>2</sub> in a cooperative manner.
- Structure:
  - Normal adult hemoglobin A (Hb A) consists of four polypeptide chains,  $\alpha_2\beta_2$ , each with its own haem group.
- When is the Hb test ordered for a patient? If anemia is suspected.
- Normal ranges:
  - Men 13-17 g/dl
  - Women 12-15 g/dl



**Table 2.3** Normal haemoglobins in adult blood.

	<b>Hb A</b>	<b>Hb F</b>	<b>Hb A<sub>2</sub></b>
Structure	$\alpha_2\beta_2$	$\alpha_2\gamma_2$	$\alpha_2\delta_2$
Normal (%)	96–98	0.5–0.8	1.5–3.2

# Haemoglobinometry

- The hemoglobin concentration (Hb) of a solution may be estimated by measurement of:
  1. It's color.
  2. Determination of its power of combining with oxygen or carbon monoxide.
  3. Analysis of its iron content.
- Two methods are used to measure hemoglobin concentration using a spectrometer:
  1. Haemoglobinocyanide (HiCN; cyanmethaemoglobin) method.
    - The most commonly used method currently.
  2. Oxyhaemoglobin (HbO<sub>2</sub>) method.
    - Used in point-of-care instruments

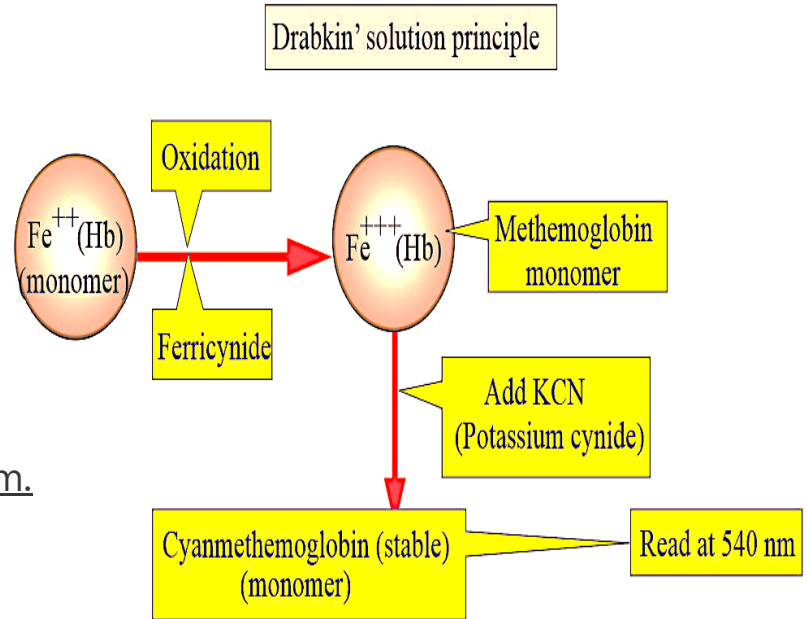
They are color or light-intensity matching techniques.

# HAEMIGLOBINCYANIDE (CYANMETHAEMOGLOBIN) METHOD

- The haemiglobincyanide (cyanmethaemoglobin) method is the internationally recommended method for determining the haemoglobin concentration of blood.
- The basis of the method is diluting the blood in **Drapkin's solution, which** contains potassium cyanide and potassium ferricyanide.
- The absorbance of the converted cyanhaemoglobin (HiCN) in the solution is then measured in a spectrometer at a wavelength of 540 nm.

# Haemiglobincyanide Principle

1. When **blood** is mixed with **Drabkin's solution**, the **potassium ferricyanide** oxidizes the **iron** to form **methemoglobin**.
2. The **potassium cyanide** then combines with **methemoglobin** → to form **cyanmethemoglobin**.
  - **Cyanmethemoglobin** is a stable color pigment read photometrically at a wavelength of 540 nm.

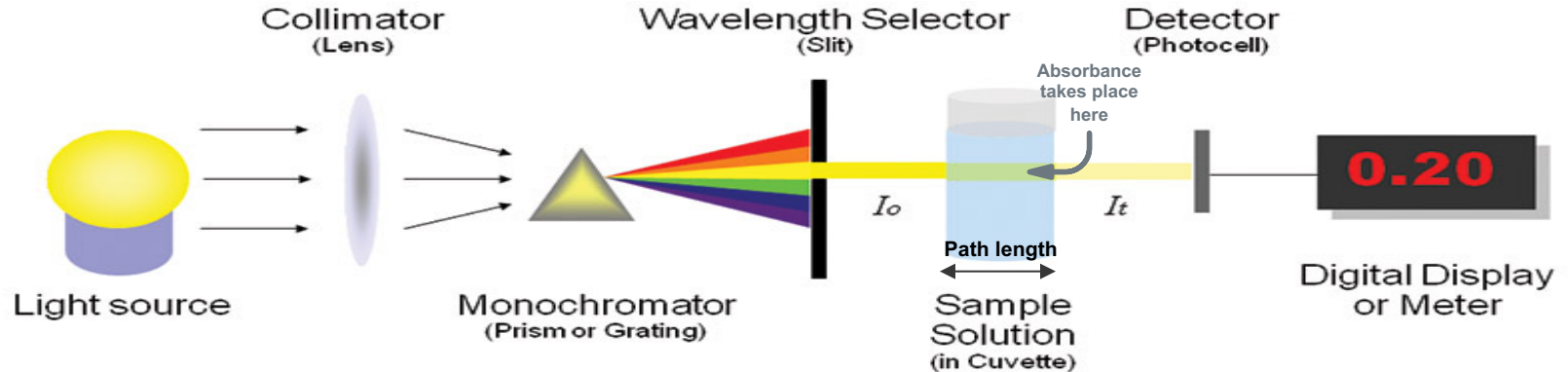


# Spectrophotometry

- A spectrophotometer is a device used to measure light intensity and can measure either the absorbance or transmittance of light.

## Principle:

- A small beam of light with a specific wavelength is emitted from the spectrophotometer, which goes through the sample in a small container called a **cuvette**.
- The spectrophotometer measures how much light is absorbed by the sample or how much of the light passes through the sample, which is transmittance.





# Haemiglobincyanide Material

- **Material:**
  - 10 or 5 ml graduated pipette.
  - P20 or P100 micropipette.
  - Tips.
  - Plastic tube.
  - Rack.
- **Sample**
  - EDTA whole blood sample.
- **Reagent:**
  - Diluent of Drapkin's solution

## DRABKIN-TYPE REAGENT

Reagent	Amount
Potassium ferricyanide (0.607 mmol/l)	200 mg
Potassium cyanide (0.768 mmol/l)	50 mg
Potassium dihydrogen phosphate (1.029 mmol/l)	140 mg
Nonionic detergent*	1 ml
Distilled or deionised water	To 1 litre

# Haemiglobincyanide Procedure

## Procedure:

1. In a plastic tube, add 4 ml of the diluent.
2. Invert the whole EDTA blood several times.
3. Aspirate 20 ul (by Micropipette ) of the blood sample and add it to the 4 ml of diluent, making the dilution 1/201.
4. Close the tube and mix by inverting the tube 3 times.
5. incubate at room temperature for 5 min. (why do we need to incubate the mixture?)
6. Pour the mixture into a cuvette and read the absorbance using a spectrometer at 540 nm against Drabkin's solution blank.

# Haemiglobincyanide Procedure

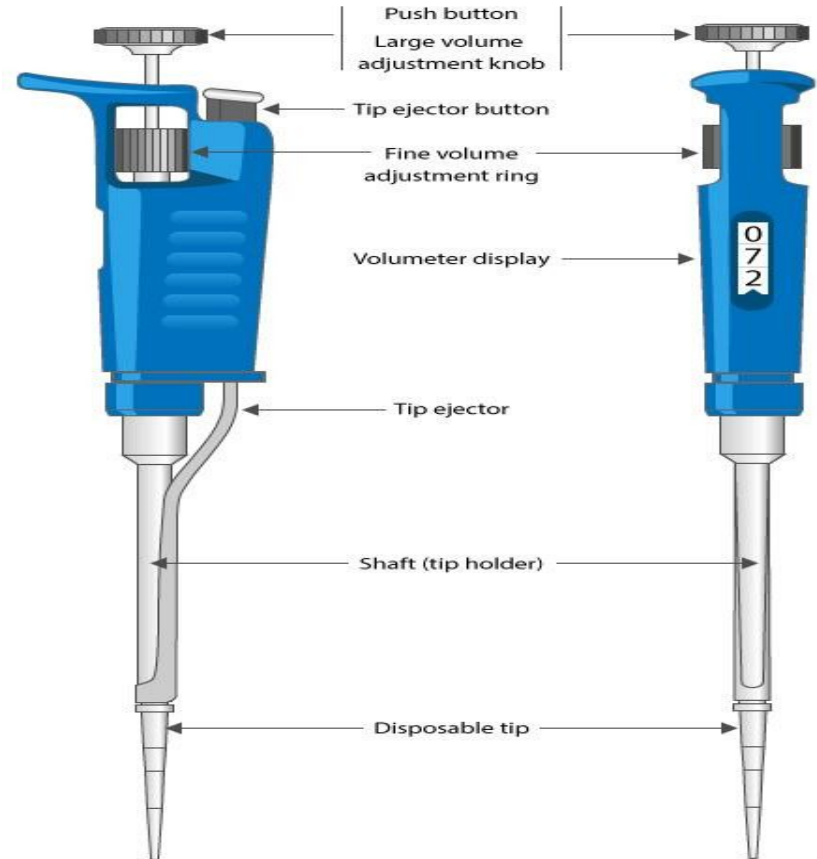
## Spectrophotometry:

1. Set the wavelength at 540 nm.
2. Start by blanking using (setting the spectrophotometer set to zero absorbance at 540nm) by using Drabkin's solution as blank.
3. Read the absorbance of your sample and record it.
4. Do your calculation.
5. The optical density (or absorbance) is proportional to the concentration of hemoglobin.

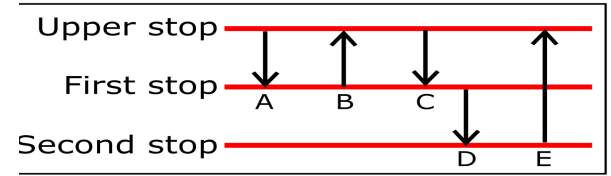
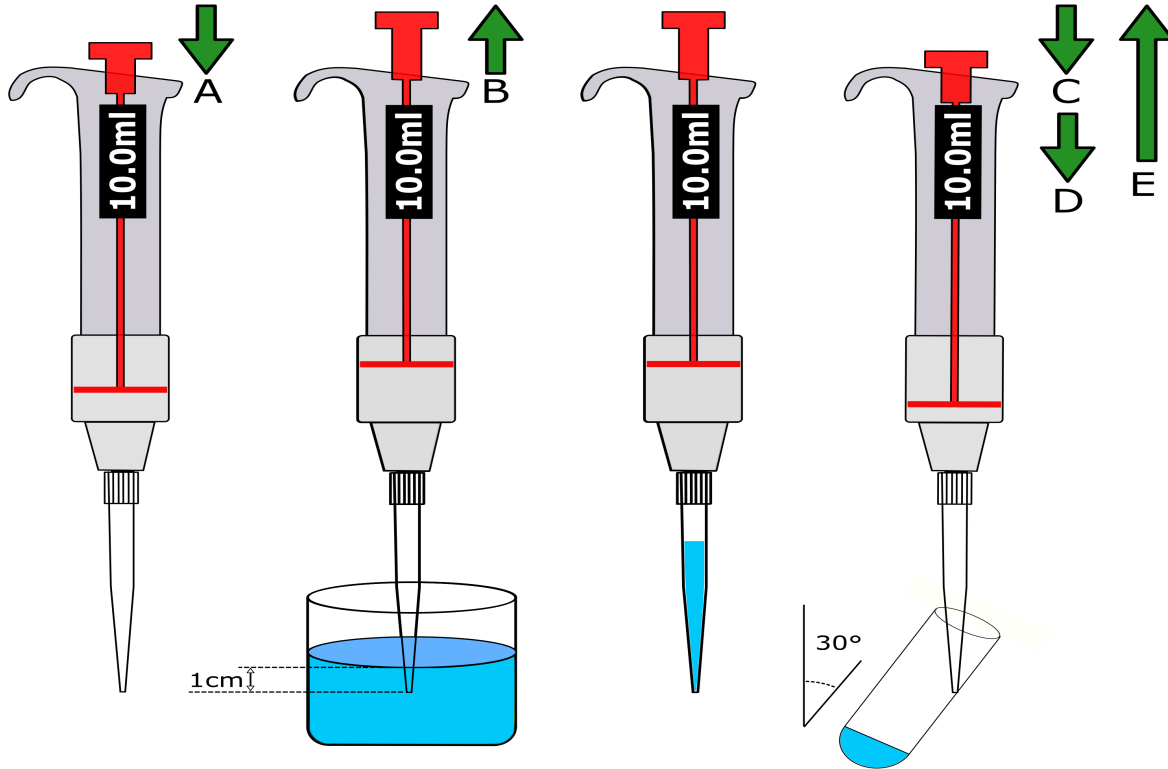
# Using Micropipettes

- Micropipettes allow you to measure and dispense small and accurate volumes of liquid solutions.
- A microliter is 1 millionth of a liter or  $10^{-6}$  L

Micropipette Model	Volume Range
P10	0.5 -10 ul
P20	2- 20 ul
P100	10 - 100 ul
P200	20 -200 ul
P1000	100 – 1000 ul



# Using Micropipettes



# Why do we need a Blank?

- In order to measure the absorbance of a particular substance in a reaction mixture, it is necessary to first “zero out” the spectrophotometer such that **only the absorbance of the substance of interest is measured.**
- This is done with a blank which is a cuvette that contains all the carrier solvents (reagents) **EXCEPT the substance of interest (analyte).**

# Result calculation

- Use the following formula:

$$\text{Hb in g/dl} = \frac{A \times 64500 \times DIL}{44 \times D \times 1000 \times 10}$$

- =  $A \times 29.3$
  
- A = Reading of absorbance of Hb solution 64500 = Molecular wt. of Hb
- 44 = Millimolar coefficient extinction
- D = Thickness of cuvette = 1
- 1000 = Conversion factor of mg to g
- DIL = Dilution Factor = 201 (total volume (solvent+ solute)/ solute)
- 10= to convert from g/l to g/dl

# Result Interpretation

- The result is expressed as g/dl and should know if below, within, or above the normal range.
- Normal ranges:
  - Men 13-17 g/dl
  - Women 12-15 g/dl
- Low Hemoglobin Level indicates:
  - Anemia (of several causes).
- High Hemoglobin Level indicates:
  - Polycythemia.
  - Smoking.



# Advantages and Disadvantages of Haemiglobincyanide

- **Advantages:**
  - Convenient for sample batching, as reading is not needed to be done immediately after dilution.
  - All forms of hemoglobin, except SHb, are readily converted to HiCN.
- **Disadvantage:**
  - The use of cyanide compounds which are considered hazardous.

## Potassium Cyanide



White, crystalline powder with a bitter almond odor. Corrosive, causes severe burns to eyes/skin/respiratory tract. Poison! Inhalation can cause anoxia, quickly leading to severe central nervous system effects. Use only with adequate ventilation.

CAS No. 0151-50-8

# Automated Hemoglobin Measurement



- Point of care testing



- Complete blood count (CBC)