

# Course Specifications (Postgraduate Degree)

<b>Course Title:</b>	Chemical Separation and Chromatographic Methods
<b>Course Code:</b>	458
Program:	Bachelor's Degree
Department:	Chemistry
College:	Science
Institution:	King Saud University







# A. Table of Contents

A. Course Identification	
B. Course Objectives and Learning Outcomes	
1. Course Description	3
2. Course Main Objective	3
3. Course Learning Outcomes	4
C. Course Content	
D. Teaching and Assessment5	
1. Alignment of Course Learning Outcomes with Teaching Strategies and Assessment Methods	5
2. Assessment Tasks for Students	6
E. Student Academic Counseling and Support6	
F. Learning Resources and Facilities6	
1. Learning Resources	6
2. Educational and research Facilities and Equipment Required	6
G. Course Quality Evaluation7	
H. Specification Approval Data7	

# A. Course Identification

1. Credit hours:	2 credit hours		
2. Course type			
🛛 Rec	quired	Γ	] Elective
3. Level/year at wh	hich this course i	s offered:	7 <sup>th</sup> or 8 <sup>th</sup> levels / 4 <sup>th</sup> year
4. Pre-requisites for	or this course (if a	any):	
352 Chem. Spectro	scopic Analytical	Methods &	353 Chem. Electrochemical Methods
5. Co-requisites fo	r this course (if a	ny):	
No requests			

### **6. Mode of Instruction** (mark all that apply)

No	Mode of Instruction	<b>Contact Hours</b>	Percentage
1	Traditional classroom	15	35
2	Blended		
3	E-learning		
4	Distance learning		
5	Other (practical in laboratory)	28	65

### 7. Actual Learning Hours (based on academic semester)

No	Activity	Learning Hours
1	Lecture	6
2	Laboratory/Studio	2
3	Seminars	
4	Others (specify)	
Total		8

### **B.** Course Objectives and Learning Outcomes

### **1.** Course Description

This course designed and conducted to the related laboratory experiments to give the students the basic principles of the extraction and separation methods. This course covers several traditional and instrumental techniques in separation and chromatographic methods of analysis.

### **2.** Course Main Objective

The main purpose of this course is to help the students to learn and understand several concepts in extraction and separation science. By the end of this course, students expected to:

-Recognize the principles of separation methods

- -Be familiar with the separation methods concepts, especially GC and HPLC techniques
- -Know the proper separation tool for specific compounds and mixtures

-Learn how to treat with experimental data

-Recognize the specific factors influencing the separation techniques

-To familiarize students with the different chromatographic techniques used in separation of chemical mixtures

-To develop the awareness in the students about the solvent extraction methods

### **3.** Course Learning Outcomes

	Aligned PLOs*	
1	Knowledge and Understanding	
1.1	Recognize the principles of separation methods	K2
1.2	Describe the differences between traditional and instrumental separation techniques	K1
1.3	Outline the most important extraction and separation methods and recognize their principles	K3
1.4	Recall knowledge of the fundamental aspects of the chromatographic techniques	K3
1.5	Define the most useful separation techniques such as GC and HPLC	K3
1.6	Describe the mechanism of separation and elution modes	K1
1.7	Define and describe the most important parameters in the chromatograms resulted from the separation technique	K3
1.8	List the most important application areas for GC and HPLC	K3
1.9	Name the most suitable separation method for specific compounds and matrices	K2
2	Skills	
2.1	Compare and differentiate between separation techniques	S1
2.2	Summarize and diagram the most important components of GC and HPLC systems	<b>S</b> 2
2.3	Calculate the most important parameters extracted from separation process	S1
2.4	Interpret and evaluate chromatograms resulted from the separation techniques	S2
2.5	Justify the proper separation tool for specific compounds	S3
3	Values	
3.1	Work independently and in group	V2
3.2	Demonstrate and analyze separation problems with group	V1
3.3	Communicate ideas to other students in the laboratory	V2
3.4	Use standard laboratory equipment and modern instrumentation	V3
3.5	Show leadership and creativity skills	V2
3.6	Use of computer and internet to search about the required information	V1
3.8	Demonstrate good and safe handling of laboratory chemicals, glassware and equipment during experiments	V3

\* Program Learning Outcomes

# **C. Course Content**

No	List of Topics	Contact Hours
1	Introduction for separation methods	1
2	Extraction and traditional separation methods	5
3	<ul> <li>-An introduction to chromatographic separations</li> <li>-Main fundamentals in chromatography methods</li> <li>-Classification of chromatographic methods</li> <li>-Chromatographic properties and evaluation</li> <li>-Applications of chromatography</li> </ul>	3
4	High performance liquid chromatography	3
5	Gas chromatography	2
6	Data analysis	1

Total	15

# **D.** Teaching and Assessment

1. Alignment of Course Learning	<b>Outcomes with</b>	<b>Teaching Strategies</b>	and Assessment
Methods			

Code	Course Learning Outcomes	Teaching Strategies	Assessment Methods
1	Knowledge and Understanding		
1.1	Recognize the principles of separation methods		
1.2	Describe the differences between traditional and instrumental separation techniques		
1.3	Outline the most important extraction methods and recognize their principles	T .	
1.4	Recall knowledge of the fundamental aspects of the chromatographic techniques	-Lecture -Discussion	-Quizzes -Direct questions
1.5	Define the most useful separation techniques such as GC and HPLC	-Homework -Training and	-Homework assignments
1.6	Describe the mechanism of separation and elution modes	-Laboratory experiments	-Midterm exam -Final exam
1.7	Define and describe the most important parameters in the chromatograms resulted from the separation technique	-Kear examples	
1.8	List the most important application areas for GC and HPLC		
1.9	Name the most suitable separation method for specific compounds and matrices		
2	Skills		
2.1	Compare and differentiate between separation techniques		
2.2	Summarize and diagram the most important components of GC and HPLC systems	-Discussion -Homework -Solving problems	-Quizzes -Direct questions -Homework
2.3	Calculate the most important parameters extracted from separation process	-Laboratory experiments -Connect of the	assignments -Midterm exam
2.4	Interpret and evaluate chromatograms resulted from the separation technique	knowledge with the real examples	-Laboratory reports -Final exam
2.5	Justify the proper separation tool for specific compounds		
3	Values		
3.1	Work independently and in group	-Solving problems with	
3.2	Demonstrate and analyze separation problems with group	group -Discussion	-Evaluating individual works
3.3	Communicate ideas to other students in the laboratory	-Writing laboratory reports	-Homework assignments
3.4	Use standard laboratory equipment and modern instrumentation	-Provide these programs for students	-Evaluating laboratory reports
3.5	Show leadership and creativity skills	-Training the students	-Demonstration
3.6	Use of computer and internet to search about the required information	-Encourage students to collect information	-Short problems and practical exams
3.7	Use of computer programs such as Microsoft Excel or other mathematical tool	through university provided Wi-Fi	F

to deal with the chromatographic parameters and evaluations-Perform experiments individually and in groups3.8Demonstrate good and safe handling of laboratory chemicals, glassware and-Perform experiments individually and in groups	Code	Course Learning Outcomes	Teaching Strategies	Assessment Methods
a main manual distribution and manufacture of the second	3.8	to deal with the chromatographic parameters and evaluations Demonstrate good and safe handling of laboratory chemicals, glassware and	-Perform laboratory experiments individually and in groups	

### 2. Assessment Tasks for Students

#	Assessment task*	Week Due	Percentage of Total Assessment Score
1	Midterm exam	6	20
2	Assignment, research, discussion and homeworks	6-15	10
3	Laboratory reports, quizzes and practical exams		30
4	Final Exam	16-17	40
5			

\*Assessment task (i.e., written test, oral test, oral presentation, group project, essay, etc.)

### E. Student Academic Counseling and Support

Arrangements for availability of faculty and teaching staff for individual student consultations and academic advice:

-Office hours: 4 h/week

-Laboratory assistance

### **F. Learning Resources and Facilities**

### **1.** Learning Resources

Required Textbooks	Ibrahim Al-Zamil, "Analytical Chemistry, Instrumental Analysis" 5 <sup>th</sup> Ed., Al-Khrigi Publisher, 2015.
Essential Reference Materials	None.
Electronic Materials	-Web of Science, Elsevier Academic Press. -Encyclopedia of chemistry and chromatography.
Other Learning Materials	<ul> <li>-All lecture notes are available on the web site of the course teacher.</li> <li>-Gary D. Christian, Purnendu K. Dasgupta, Kevin A. Schug, Analytical Chemistry, 7th ed., 2013, Wiley, USA.</li> <li>-Douglas A. Skoog, F. James Holler, Stanley R. Crouch, Principles of Instrumental Analysis, 7th ed., 2018, Cengage Learning, USA.</li> <li>-Kevin Robards, Paul R. Haddad, Peter E. Jackson, Principles and Practice of Modern Chromatographic Methods, 2012, Elsevier Academic Press, Netherlands.</li> </ul>

### 2. Educational and research Facilities and Equipment Required

Item	Resources		
Accommodation (Classrooms, laboratories, demonstration rooms/labs, etc.)	<ul> <li>-Lecture hall for 30 students equipped with modern teaching technology (projector, smart board, computer and internet)</li> <li>-Laboratory should be supplied with the related chemicals, glassware and the required instruments</li> <li>-Laboratory in accordance with the rules of safety and body protection accessories should be available to all students</li> </ul>		

Item	Resources		
<b>Technology Resources</b> (AV, data show, Smart Board, software, etc.)	<ul> <li>The presence of computer, smart board and internet access in the in classrooms</li> <li>E-podium &amp; data show available in all lecture room</li> </ul>		
Other Resources (Specify, e.g. if specific laboratory equipment is required, list requirements or attach a list)	-The presence of chemicals and standards used in analytical experiments -The presence of related analytical equipment and instruments such as GC, HPLC, separation columns and accessories, pH meter, analytical balance,etc.		

## **G.** Course Quality Evaluation

Evaluation Areas/Issues	Evaluators	<b>Evaluation Methods</b>
Feedback on Effectiveness of	Student and post	survey
Teaching	graduate student	Survey
Teaching	Departmental council and analytical group	- Discussion and meetings
Processes for Improvement of Teaching	Faculty and department	direct
Extent of achievement of course learning outcomes,	Department	direct

**Evaluation Areas/Issues** (e.g., Effectiveness of teaching and assessment, Extent of achievement of course learning outcomes, Quality of learning resources, etc.)

**Evaluators** (Students, Faculty, Program Leaders, Peer Reviewer, Others (specify) Assessment Methods (Direct, Indirect)

# **H. Specification Approval Data**

Council / Committee	
Reference No.	
Date	