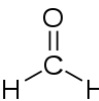
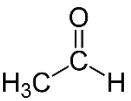
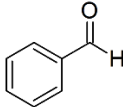
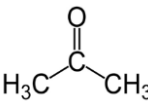


Aldehyde and Ketone

	Aldehyde			Ketone	
	Name	Formaldehyde	Acetaldehyde	Benzaldehyde	Acetone
	Mol. Formula	CH ₂ O	C ₂ H ₄ O	C ₇ H ₆ O	C ₃ H ₆ O
	Mol. Structure				
	Physical state	Colorless - Liquid=gas	Colorless -Liquid	Colorless -Liquid	Colorless - Liquid
Carbonyl group	Nickelophilic addition 2,4DNPH 1ml of Aldehyde or Ketone+ 2,4DNPH <i>(But Ketone before add 2,4DNPH mixed with water)</i>	+Ve result Yellow-orange Precipitate	+Ve result Yellow-orange Precipitate	+Ve result Yellow-orange Precipitate	+Ve result Yellow-orange Precipitate
Oxidation and Reduction	1-KMnO₄ 1ml of Aldehyde +NaOH+Δ	+Ve result disappearance of the purple color of KMnO ₄ appearance of a brown precipitate	+Ve result disappearance of the purple color of KMnO ₄ appearance of a brown precipitate	+Ve result disappearance of the purple color of KMnO ₄ appearance of a brown precipitate	- Ve result
	2-Fehling,s (A+B) 1ml of Aldehyde +Fehling,s + Δ	+Ve result Red ppt of copper oxide appears	+Ve result Red ppt of copper oxide appears	+Ve result Red ppt of copper oxide appears (Long time)	- Ve result
	3-Ammonical silver Nitrate (Tollen,s) drop of silver Nitrate+ 10% NaOH black ppt+ NH ₄ OH add to of Aldehyde +Δ	+Ve result Silver mirror	+Ve result Silver mirror	+Ve result Silver mirror	- Ve result
Formaldehyde	Salicylic acid 1ml of Formaldehyde + Small amount of Salicylic acid +drop of conc H ₂ SO ₄ carefully in side the wall of tube	+ Ve result Red color	- Ve result	- Ve result	- Ve result