# Chapter 10 Polymer Characteristics

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# Stress – Strain Behavior (I)

 The description of stress-strain behavior is similar to that of metals



Polymers can be brittle (A), plastic (B), or highly elastic, or elastomer (C). Deformation shown by curve C is totally elastic (rubber-like elasticity, large recoverable strain at low stress levels).

### Stress – Strain Behavior (II)

- Characteristics of stress-strain behavior:
- ✓ Modulus of elasticity defined as for metals
- ✓ Ductility (%EL) defined as for metals
- Yield strength For plastic polymers (B), yield strength is defined by the maximum on curve just after the elastic region (different from metals)
- Tensile strength is defined at the fracture point and can be lower than the yield strength (different from metals).



# Stress – Strain Behavior (III)

- Moduli of elasticity for polymers are ~ 10 MPa 4 Gpa (compare to metals ~ 50 - 400 GPa)
- Tensile strengths are ~ 10 100 MPa (compare to metals, hundreds of MPa to several GPa)
- Elongation can be up to 1000 % in some cases (< 100% for metals)

Mechanical properties of polymers change dramatically with temperature, going from glass-like brittle behavior at low temperatures to a rubber-like behavior at high temperatures.
 Polymers are also very sensitive to the rate of deformation (strain rate). Decreasing rate of deformation has the same effect as increasing T.

#### Stress – Strain Behavior (IV)

Temperature increase leads to:

- Decrease in elastic modulus
- Reduction in tensile strength
- Increase in ductility

The glass transition temperature (Tg) of PMMA ranges from 85 to 165 °C – all of the above curves are for temperatures below Tg.



- Amorphous polymer: glass at low temperatures, rubber at intermediate temperatures, viscous liquid at high T.
- Low temperatures: elastic deformation at small strains (σ = Eε). Deformation is instantaneous when load is applied. Deformation is reversible.
- High temperatures: **viscous behavior. Deformation is** time dependent and not reversible.
- Intermediate temperatures: **viscoelastic behavior.** Instantaneous elastic strain followed by viscous time dependent strain.
- Viscoelastic behavior is determined by rate of strain (elastic for rapidly applied stress, viscous for slowly applied stress)





Rate dependence of viscoelastic properties in a silicone polymer (Silly Putty). Picture by Geon Corp.



Load is applied at  $t_a$  and released at  $t_r$ 

- Viscoelasticity can be characterized by the viscoelastic relaxation modulus:
- ✓ Sample is strained rapidly to predetermined strain
- Stress required to maintain this strain
  ε0 over time is measured at constant T.
- ✓ Stress decreases with time due to molecular relaxation processes.
- $\checkmark$  Relaxation modulus can be defined as

 $E_r(t) = \sigma(t)/\epsilon_0$ 

E<sub>r</sub>(t) is also a function of temperature



To show the influence of temperature, the relaxation modulus can be plotted at a fixed time for different T:



Temperature dependence for different polymer structures A : Largely crystalline isotactic polystyrene. Glass transition region limited – small amount of amorphous material B: Lightly cross-linked atactic polystyrene – leathery region extends to decomposition temperature: no melting C: Amorphous polystyrene



#### **Factors that Influence Mechanical properties**

- Temperature and strain rate (already discussed)
- Chain entanglement, strong intermolecular bonding (van der Waals, cross-links) increase strength
- Drawing, analog of work hardening in metals, corresponds to the neck extension. Is used in production of fibers and films. Molecular chains become highly oriented ⇒ properties of drawn material are anisotropic (perpendicular to the chain alignment direction strength is reduced)
- Heat treatment changes in crystallite size and order
  - o **undrawn material**: Increasing annealing temperature leads to
    - ✓ increase in elastic modulus
    - ✓ increase in yield/tensile strength
    - ✓ decrease in ductility

Note that these changes are opposite from metals

 drawn material: opposite changes (due to recrystallization and loss of chain orientation)

#### **Factors that Influence Mechanical properties**

- Tensile strength increases with molecular weight effect of entanglement
- Higher degree of crystallinity stronger secondary bonding stronger and more brittle material



# Polymerization

- Polymerization is the synthesis of high polymers from raw materials like oil or coal. It may occur by:
- Addition (chain-reaction) polymerization, where monomer units are attached one at a time (discussed in previous Chapter). Has three distinct stages: initiation, propagation, and termination.
- 2. Condensation (step reaction) polymerization, by stepwise intermolecular chemical reactions that produce the mer units.
  - ✓ Usually there is small by-product that is then eliminated
  - ✓ Significantly slower than addition polymerization
  - Often form trifunctional molecules that can form crosslinked and network polymers

