

جامعة الملك سعود - كلية العلوم - قسم الكيمياء الفصل الصيفي 1438 هـ (2017 م) الاختبار الفصلّي الأولَ البديلُ في مقرر 145 كيم (25-11-1438 هـ) الزمن: 90 دقيقة

رقم الطالب:

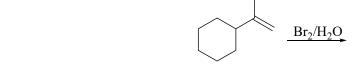
أسم الطالب: أستاد المقرر أ. د. /

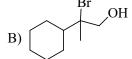
نموذج الأجابة: ملاحظة هامة: تصحيح الامتحان سيكون بناء على الإجابة المكتوبة في الجدول أسفل (حرف الإجابة الصحيحة) ولن ينظر الى بقية الأوراق والتي تعتبر مسودة.

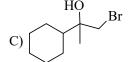
الإجابة	رقم السؤال	الإجابة	رقم السؤال
С	16	C	1
С	17	D	2
В	18	С	3
A	19	A	4
D	20	В	5
С	21	D	6
A	22	A	7
В	23	С	8
D	24	D	9
A	25	В	10
С	26	A	11
A	27	С	12
D	28	В	13
В	29	D	14
A	30	A	15

Choose the correct answer for each of the following Questions

1. The product of the reaction shown below is







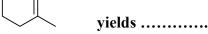
- 2. The number of structural isomers for the molecular formula C_5H_{12} is
 - A) 1

B) 2

C) 4

D) 3

3. Ozonoloysis of



$$D) \bigvee_{Q} Q$$

4. The product of the following reaction is

$$H_2$$
 / Pd

A) CH₃CH₂CH₃

B) CH₃CH₃

- pd | D) CH₃ -CH-CH₃
- 5. The type of hybridization of the selected carbon in the following structure is ...



A) sp^3d

- B) sp^2
- C) sp^3

D) sp

6. The compound \underline{X} in the following reaction is

$$\mathbf{X} \xrightarrow{\text{H}_2\text{O}, \text{H}_2\text{SO}_4, \text{HgSO}_4} \overset{\text{O}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{\text{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}}{\overset{|}}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}}{\overset{|}}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}{\overset{|}}$$

- A) Butane
- B) Butene
- C) 2-Chlorobutene
- D) ButYne
- 7. The carbon bearing a negative charge is called
 - A) carbanion
- B) Free radical
- C) carbocation
- D) cation
- 8. At room temperature, alkenes from C₅ to C₁₇ are
 - A) gases

- B) solids
- C) liquids
- D) semisolids
- 9. The compound with the highest boiling point is
 - A) n-Pentane

B) 2,2-Dimethylbutane

C) 2-Methylpentane

- D) *n*-Hexane
- 10. The product of the following reaction is

11. The geometry of the selected carbon is

- A) Tetrahedral
- B) Trigonal planar
- C) Bent
- D) Linear

- 12. The number of bonds in 1-pentyne are
 - **A)** Twelve σ and four π bonds

B) Twelve σ and three π bonds

C) Twelve σ and two π bonds

- **D)** Twelve σ and one π bonds
- 13. Acetylene is the common name for?
 - A) Propene
- B) Ethyne
- C) Ethene
- D) Ethane

14. The reagent X needed for the following transformation is

15. The product of the following reaction is

- A) *trans*-2-Butene
- B) *cis*-2-Butene
- C) *trans*-2-pentene
- D) Butane

16. The incorrect answer about the compound H₃C-CH=CH-CH₃ is

- A) It can exist as trans or cis form
- B) Its name is 2-butene
- C) Addition of water to this compound gives two structural isomers
- D) Its hydrogenation gives Butane

17. The IUPAC name of

- A) 1-Cyclopentyl-4-bromo-1-octyne
- B) 8-Cyclopentyl-5-bromo-8-octyne
- C) 4-Bromo-1-cyclopentyl-1-octyne
- D) 4-Bromo-1-pentyl-1-octyne

18. What is the IUPAC name for the following compound?

- A) 1,3-Pentamethylpropane
- C) 2,4,4-Trimethylpentane

- B) 2,2,4-Trimethylpentane
- D) 1,1,1,3-tetramethylbutane

19. The following name is incorrect

A) 2-ethylpropane

B) 2,2-dimethylpentane

C) 2-methylhexane

D) *n*-pentane

20. Select the correct formula of 2-methyl pentane?					
A) C_5H_{12}	B) C ₅ H ₁₆	C) C ₆ H ₁₂	D) C ₆ H ₁₄		
21. Which of the following is the correct IUPAC name of the following compound?					
A) <i>E</i> -2-Ethyl-2-pentene C) <i>Z</i> -3-Methyl-3-hexene		B) <i>Z</i> -2-Ethyl-2-pentene D) <i>E</i> -3-Methyl-3-hexene			
22. What is the reagent needed for the following transformation?					
	Br	? //			
	Br				
A) Zn/AcOH	B) H ₃ O ⁺	C) KOH/Alcohol/Heat	D) H ⁺ /Heat		
23. Which one of the following is a nucleophile?					
A) AlCl ₃	B) CN	C) H ₃ O ⊕	D) BF ₃		
24. Which of the following molecules has an ionic bond?					
A) NH ₃	B) H ₂	C) CH ₄	D) NaCl		
25. The monochlorinated ethane can be obtained under the following experimental conditions is					
$A)H_3C-CH_3 + C$	Cl ₂ UV light	B)H ₃ C-CH ₃ (excess) +	Cl ₂ UV light		
$C)H_3C-CH_3 + C$	Dark, R.T.	$D)H_3C-CH_3 + Cl_2 (exc$	ess) UV light		
26. When carbon is be	onded to two other	r carbon atoms, it is called a	a		
A) Primary carbonC) Secondary carbon		B) Tertiary carbonD) Quaternary carbon			

B) Chloroethane

D) Chloroethyne

27. The IUPAC name for Vinyl chloride is

A) Chloroethene

C) Chlorobenzene

28. The product of the following reaction $CH_2Br = \frac{1) \text{ Mg/dry ether}}{2) H_3O^+}$ is ...

- A) Pentane.
- B) Propane.
- C) *n*-Butane
- D) Isobutane.

29. The IUPAC name for the following formula is

$$\begin{array}{c} \mathsf{H_3C} \\ \mathsf{H_3C} \\ \mathsf{H_3C} \\ \mathsf{CH_2} \\ \mathsf{H_3C} \end{array} \\ \mathsf{CH_2} \\ \mathsf{H_3C} \\ \end{array}$$

- A) 5-Ethyl-6,6-dimethylheptane
- B) 3-Ethyl-2,2-dimethylheptane
- C) 2,2-Dimethyl-3-ethylheptane
- D) 6,6-Dimethyl-5-ethylheptane

30. Sodium reacts with alkyl halides in dry ether to form alkanes, the reaction is known as

A) Wurtz synthesis

B) Grignard reagent

C) nucleophilic substitution

D) none