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Journal of King Saud University – Science

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Original article

Eco-friendly synthesis of antibacterial zinc nanoparticles using *Sesamum indicum* L. extract



Sara Zafar^a, Asma Ashraf^b, Muhammad Umar Ijaz^c, Saima Muzammil^d, Muhammad Hussnain Siddique^{e,*}, Sidra Afzal^a, Rahat Andleeb^b, Khalid A. Al-Ghanim^f, Fahad Al-Misned^f, Zubair Ahmed^f, Shahid Mahboob^{f,*}

^a Department of Botany, Government College University, Faisalabad, Pakistan

^b Department of Zoology, Government College University, Faisalabad, Pakistan

^c Department of Zoology, Wildlife and Fisheries University of Agriculture, Faisalabad, Pakistan

^d Department of Microbiology, Government College University, Faisalabad, Pakistan

^e Department of Bioinformatics and Biotechnology, Government College University, Faisalabad, Pakistan

^f Department of Zoology, College of Science, King Saud University, Riyadh, Saudi Arabia

ARTICLE INFO

Article history: Received 20 September 2019 Revised 15 October 2019 Accepted 29 October 2019 Available online 13 November 2019

Keywords: Zinc nanoparticles Characterization Antibacterial activity FTIR XRD

ABSTRACT

The present study was conducted to prepare cost effective and environment friendly zinc nanoparticles (Zn NPs) by using ZnSO₄·7H₂O and extract of sesame seeds. Zn NPs were characterized by using scanning electron microscope (SEM), Fourier-transform infrared spectroscopy (FTIR), X-Ray Diffraction and antibacterial potential was checked by using disc diffusion method. Solution colour turned yellow which ultimately confirmed the formation of Zn NPs and seed extract was used as a reducing agent. Crystalline size of the NPs was revealed by XRD spectra i.e. 9.07 nm. Functional groups were identified by the FTIR. SEM analysis given idea about the spherical shape of NPs. Antibacterial activity of the Zn NPs against different bacteria (*S. pneumoniae, P. aeruginosa, E. coli* and *S. aureus*) was measured by inhibition zone which ranged from 3.3 to 12.6 mm.

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1. Introduction

Nanotechnology aims the production of very small sized particles that have nanoscale dimension (Agarwal et al., 2017, Fazalur-Rehman, 2018). A number of chemical and physical approaches are used to synthesize nanoparticles with desired properties but these methods are very expensive, harmful for the humans and environment, labor intensive, toxic and high amount of energy and pressure is required for their proper completion (Taheriniya and Behboodi, 2016). NPs are biologically produced by plants, algae, fungi, bacteria, diatoms and human cells. Plants are preferred among all the green synthesis routes for nanoparticles (Makarov et al., 2014, Mirzaei and Darroudi, 2017).

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Characteristics of nanoparticles as well as their stability, size and shape are greatly affected by the temperature, concentration, pH and many other factors during the formation process (Shah et al., 2015). Different kinds of NPs including gold, iron, zinc, silver, copper, platinum as well as oxides of metals and non-metals etc are formed from prokaryotes and eukaryotes (Hasan, 2015).

Among all, Zn NPs are preferred because they can generate ROS and are good semiconductors (Malaikozhundan, 2018, Srivastava et al., 2013). Characteristics of Zn NPs depend on their size, concentration and shape (Siddiqi et al., 2018). Extracts of the plants are used as bioreductant and capping agent for producing NPs (Shaik et al., 2018). Phytomolecules (Stephen and Seethalakshmi, 2013) and secondary metabolites present in plants extract play vital role during nanoparticle synthesis (Marslin et al., 2018, Mourdikoudis et al., 2018).

Sesamum indicum L. is known as Queen of oilseeds. It is rich in the natural antioxidants (Pathak et al., 2014). During green synthesis, extract of the sesame is used as reducing and gelling agent (Manikandan et al., 2016).

The aim of this study was the production of the Zn NPs by green method through bio-reduction using Sesame seed extract, and to check their antibacterial potential.

https://doi.org/10.1016/j.jksus.2019.10.017

^{*} Corresponding authors.

E-mail addresses: mhs1049@gmail.com (M.H. Siddique), shahidmahboob60@ hotmail.com (S. Mahboob).

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2. Materials and methods

2.1. Seeds extraction

Sesame seeds were washed using distilled water, dried and then grinded. The seed powder (20 g) was mixed with the distilled water (100 ml), boiled for 45 min at 75 °C (Saravanakkumar et al., 2016). Extract was collected and stored at 4 °C.

2.2. Preparation of Zn NPs

Zinc nanoparticles were prepared by adding 20 ml of the sesame seed extract in 80 ml of the 1 mM ZnSO₄·7H₂O solution. Sonicate solution for 1 h at 60 °C. A change in the color to yellow indicated the formation of Zn NPs (Eya'ane Meva et al., 2016).

2.3. Characterization

Diameter of crystalline Zn NPs was determined by using X-Ray diffraction by Scherrer's equation i.e.

$t = k\lambda/\beta \cos\theta$

where t = size of the crystallite, k = shape factor, λ = wavelength of radiation.

2.4. SEM and FTIR

SEM was used for scanning surface of sample. Qualitative and quantitative examination as well as functional groups in the sample was determined by Fourier Transform Infrared Spectroscopy (FTIR) (Eya'ane Meva et al., 2016).

2.5. Zeta potential

Stability of NPs was dogged by zeta potential (Saranya et al., 2017).

2.6. Antibacterial potential

Antibacterial potential of the Zn NPs was examined using Kirby-Bauer standard disc diffusion process by various dilutions (50, 100, 150 and 200 μ g/mL) of Zn NPs (Velusamy et al., 2015). The application of bacterial suspension (100 μ L) on the surface of nutrient agar plate was done overnight. Antibacterial disc (Gentamicin), discs coated with different concentrations of Zn NPs and seed extract placed on the inoculated plates then kept in incubator for 24 h at 37 °C. After that, inhibition zone was noticed around each disc.

2.7. Statistical analysis

The data was statistically analyzed by analysis of variance technique using completely randomized design in factorial arrangement.

3. Results

3.1. XRD analysis

The XRD diffraction spectrum of green synthesized Zn NPs exhibited peaks at 2θ (32.171, 34.901, 36.807, 48.631, 57.473, 63.701, 68.202) conforming to Zn (1 0 0), (0 0 2), (1 0 1), (1 1 0), (1 0 3), (2 0 0) and (0 0 4) planes of the standard XRD. The XRD design suggested that zinc nanoparticles were crystalline in the nature. The XRD pattern revealed a strong plane of zinc at 101 and other weaker planes (Fig. 1). The average size of Zn NPs was determined as 9.07 nm.

3.2. FTIR analysis

FTIR spectroscopy is an effective approach for determining the major functional groups (reduced bio-molecules) necessary for the green synthesis of NPs via the reduction and stabilization of Zn ions. The FTIR spectrum revealed a number of absorption bands (Fig. 2: Table 1). Peaks were observed at 428.73, 728.18, 1021.33. 1458.56, 1563.78, 2845.31, 2944.14 and 3425.88 cm⁻¹. At 728.18 cm⁻¹ and 1021.33 cm⁻¹ absorption band revealed the existence of C-H bending of the aromatic alkanes, alkenes, alkynes and aromatic hydrocarbons. Similarly, presence of N-H bending (Primary, secondary amines and amides) and C-H bending was confirmed by peaks at the 1458.56 cm^{-1} and 1563.78 cm^{-1} . At the peak of 2845.31 cm⁻¹ alcohol was observed with C–O vibrations. Existence of C=O functional group of carboxylic acid in addition with ketones and aldehydes was confirmed by peak at the 2944.14 cm⁻¹. Presence of the O-H vibration of phenol and alcohols was revealed by the absorption band at the 3425.88 cm⁻¹. All these vibrations showed that various biomolecules are involved in stabilization and reduction of NPs.



Fig. 1. XRD pattern of the Zn NPs.

3.3. SEM analysis

SEM analysis verified the presence of the Zn NPs created by treating zinc sulfate solution with *Sesamum indicum* seed extract. SEM revealed about average size of NPs as 70 nm as well as about their spherical shape. The shape and size of the NPs depend on extract concentration and ratio of zinc sulfate to the seed extract (Fig. 3).

3.4. Zeta potential

Surface charge of Zn NPs was determined by zeta potential. Zeta potential measured stability of NPs as -18.6 mV (Fig. 4).

3.5. Antibacterial potential

Antibacterial strength of synthesized Zn NPs was examined using Kirby-Bauer disc diffusion process. Different dilutions (50,

Table 1

FTIR result showed stretching and bending vibration of compounds with functional groups.

 Frequencies cm- ¹	Functional group	Type of vibrations
428.73	Zn-O	Stretching vibration
728.18	C-H	Bending vibration (alkanes, alkynes, alkenes and aromatic hydrocarbons)
1021.33	C-H	Bending vibration (aromatic hydrocarbons)
1458.56	C-H	Bending vibration (Primary and secondary
	N-H	amides and amines as well as aromatic hydrocarbons)
1563.78	C-H	Bending vibration (Primary and secondary
	N-H	amides and amines as well as aromatic
		hydrocarbons)
2845.31	C-0	Stretching vibration (alcohols)
2944.14	C=0	Stretching vibration (carboxylic acids, ketones and aldehydes)
3425.88	O-H	Stretching vibration (alcohols and phenols)

100, 150 and 200 μ g/mL) of the NPs were checked against *E. coli*, *P. aeruginosa*, (gram negative), *S. aureus* and *S. pneumoniae* (gram positive) (Fig. 5). Solution of 200 μ g/mL revealed the highest zone of inhibition against bacteria (Tables 2 and 3). In comparison of inhibition zone of Zn NPs, antibiotic (gentamicin) was used as a standard. Zones of inhibition were present in range of 3.3–12.6 mm. NPs cause severe damage to the target by attacking its cell membrane and cell wall and disturbing its whole functioning.

4. Discussion

Sesamum indicum seed extract was used for the preparation of green eco-friendly Zn NPs as the bio-molecules in the leaf extract are effective in stabilizing the Zn NPs (Al-Dhabi and Valan Arasu, 2018). A change in extract color with ZnSO₄ solution showed reduction in zinc ions which confirmed the formation of Zn NPs. Formation of the Zn NPs was revealed by yellow colour appearance (Bisquera et al., 2017). In the biological methods nanoparticles are formed by the reduction of metal salt, as metal salt and extract mixed together (Sirisha and Mary, 2016).

In the medicinal field Zn NPs are gaining much importance due to their strong antibacterial activity as depicted in our study (Elshama et al., 2018). Characterization of the NPs is done by using highly precise techniques i.e. XRD (Sharma et al., 2012), SEM (Hodoroaba et al., 2016), Zeta potential (Clogston and Patri, 2011) and FTIR (Devaraj et al., 2013). By using XRD analysis formation of Zn NPs was determined with average size 9.07 nm and spectral analysis showed the nature of crystallization of particles and their crystalline nature (Vijayakumar et al., 2018).

Biological molecules involved in stabilizing the Zn Nps were determined by FTIR. The main peaks showed primary and secondary amides and amines, proteins, alcohols, aromatic hydrocarbons, aldehydes, ketones and carboxylic acids as well as stretching and bending vibrations proved that all these are connected with Zn NPs (Jamdagni et al., 2018). SEM analysis revealed average size of Zn NPs i.e. 70 nm, shape as spherical and also revealed about the surface morphology (Santhoshkumar et al., 2017). SEM revealed



Fig. 2. FTIR spectrum of Zn NPs.



Fig. 3. SEM images showing spherical Zn NPs with size of 70 nm.

that formation of NPs with larger size take place because of evaporation of solvent in preparation process which ultimately led to variation in particle size. Zeta potential of the Zn NPs depicted that prepared NPs are stable (Kim et al., 2014).

Sesmum indicum seed extract mediated Zn NPs exhibited high antibacterial potential against different bacteria *P. aeruginosa* (Hoseinzadeh et al., 2014), *E. coli* (Şahin et al., 2017), *S. aureus* (Mostafa, 2015) and *S. pneumoniae* (Bhattacharyya et al., 2018). Antibacterial potential of Zn NPs was proved from the calculations of inhibition zone. Inhibition zone of 12.6 mm was considered as a largest zone (Palanikumar et al., 2014).

5. Conclusion

Present study suggested that biological method of Zn NPs synthesis is the most economical, reliable and easy way to synthesize NPs, as a green synthesis method. The characterization through XRD spectra revealed about the size and nature of NPs, the size and shape of NPs have direct relation with the concentration of seed extract and relative amount of zinc sulfate heptahydrate to extract. SEM images revealed 70 nm size along with spherical shape of NPs. FTIR revealed that bio-molecules are responsible for the stabilization and reduction of the NPs. Zeta potential

	Zeta F	otential		
		Mean(mV):	Area(%):	St Dev (mV)
Zeta-Potential (mV): -18.6	Peak 1:	-18.6	100	5.17
Zeta deviation(mV): 5.17	Peak 2:	0.000	0.0	0.000
Conductivity(mS/cm): 1.33	Peak 3:	0.000	0.0	0.000

Zeta Potential Distribution



Zeta Potential (mV)





Escherichia coli

Staphylococcus aureus



Fig. 5. AB antibiotic (Gentamicin), E (seed extract), 50, 100, 150 and 200 µg/ml are dilutions of Zn NPs).

Table 2

Inhibiting activity of Zn NPs against four different bacteria.

Bacteria		Extract	Gentamicin	50 µg/ml	100 µg/ml	150 µg/ml	200 µg/ml
Escherichia coli Pseudomonas aeruginosa Staphylococcus aureus Streptococcus pneumoniae	A B C D		13 mm 13.9 13.3 14.6	3.7 mm 4.1 -	5.6 mm 9.2 3.3 6.3	9.7 mm 9.9 3.7 8.2	10.4 mm 12.6 4.1 8.6

Table 3

Mean square values from analysis of variance for zone of inhibition produced by zinc nanoparticles against various strains of the bacteria (*Pseudomonas aeruginosa, Escherichia coli, Streptococcus pneumoniae* and *Staphylococcus aureus*).

Source	df	SS	MS	F	Р
Bacterial strains (BS)	3	170.045	56.682	1162.7009	0.000***
Treatments (T)	5	1472.28	294.456	6040.123	0.000***
Interaction					
BS \times T	15	125.65	8.37	171.829	0.000***
Error	48	2.34	0.048		
Total	71	1770.315			

dogged the stability of Zn NPs. Green synthesized Zn NPs possess greater antibacterial activity against various bacteria (*S. pneumonia, P. aeruginosa, S. aureus* and *E. coli*). This study illustrates new and easy way for green synthesized Zn NPs with numerous uses in the medicinal field.

Conflict of interest

There were no competing interests among authors.

Acknowledgements

The facilities and support provided by the Department of Botany and Zoology, Government College University Faisalabad are highly acknowledged. "The authors (KAAG, FAM and SM) express their sincere appreciation to the Researchers Supporting Project (RSP-2019-93), King Saud University, Riyadh, Saudi Arabia".

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