## Carboxylic acid

Name	Acetic acid	Benzoic acid
Mol. Formula	$C_2H_4O_2$	$C_7H_6O_2$
Mol. Structure	H₃C—COOH	
Physical state	Liquid	Sold
Solubility	Soluble (Miscible) in water and	Soluble (Miscible) in
	Acidize	NaOH and Acidize
Strong Acidize	Strong	Weak
1ml or 0.5 g of Acid+Na <sub>2</sub> CO <sub>3</sub>		
Bromophenol blue	Yellow pH=3	Yellow pH=3
Conc. H <sub>2</sub> SO <sub>4</sub>	The smell of acetic acid	
	increases	
Esterification	The smell of acetic acid	
	increases	
1-1ml of carboxylic acid +Ethanol + Conc .H <sub>2</sub> SO <sub>4</sub> +∆		
2-Add to backer contain 10 ml water +Na <sub>2</sub> CO <sub>3</sub>		

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$$CH_3COOH + CH_3CH_2OH \longrightarrow CH_3COOCH_2CH_3 + H_2O$$