

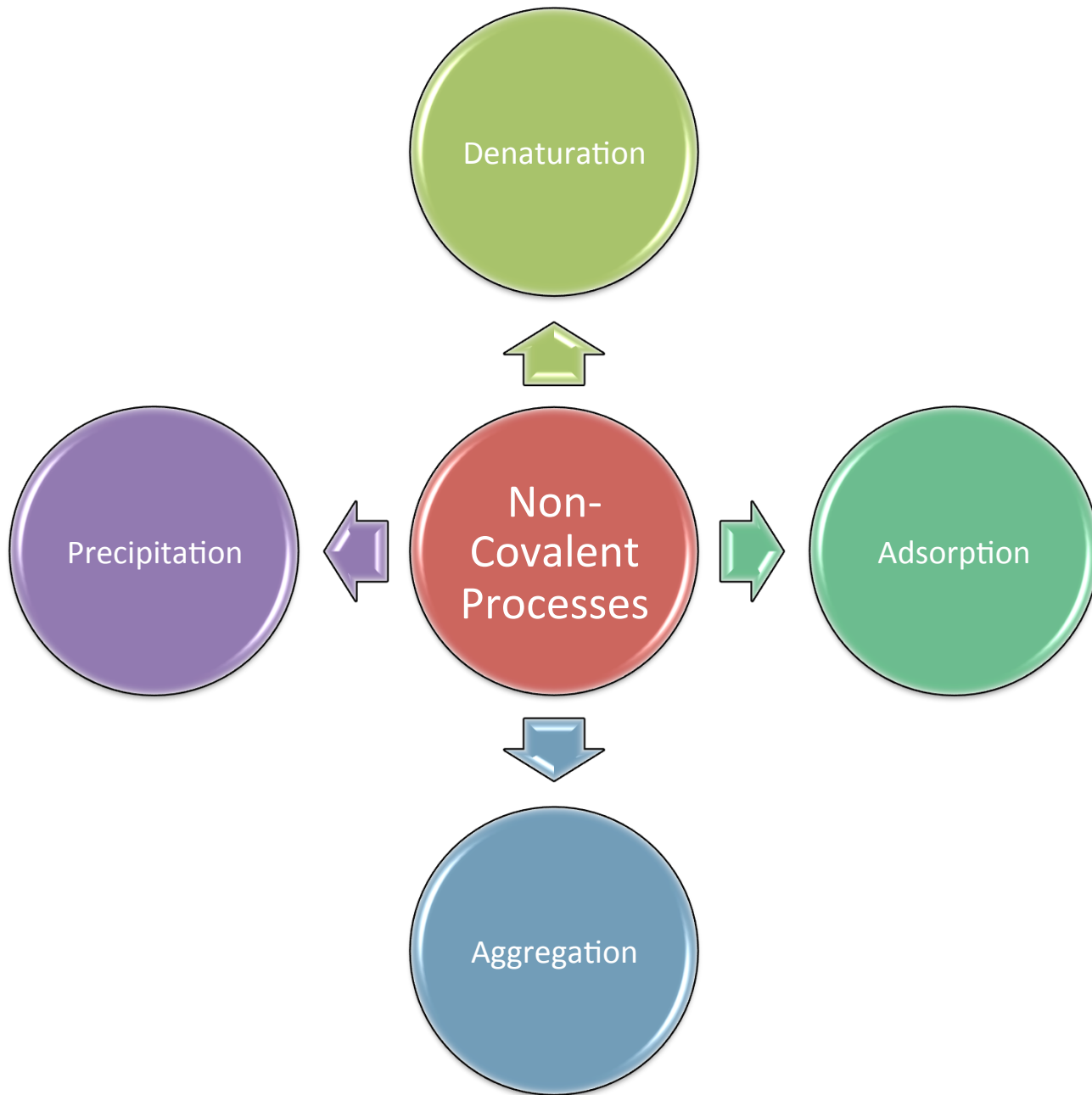
Non-Covalent Mechanisms of Protein Instability

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Objectives of this lecture

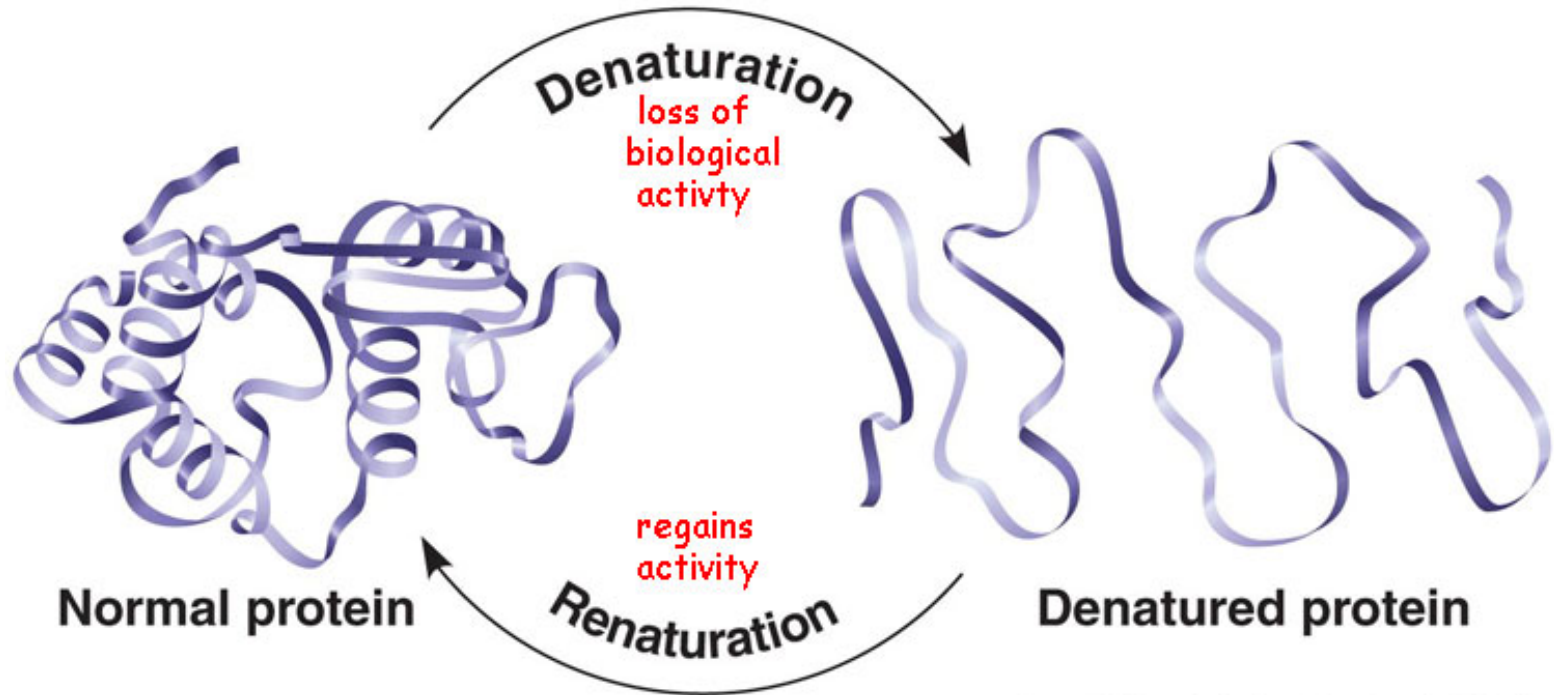
By the end of this lecture you will be able to:

1. Describe the challenges in pharmaceutical proteins production
2. Distinguish between the different mechanisms of protein instability
3. Predict the mechanism of degradation from peptide primary structure



Denaturation (unfolding)

agents: pH, temp, ionic strength, solubility



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- Hi or low pH (>10 , <3)
- Hi or low temp ($>60^{\circ}\text{C}$, $<0^{\circ}\text{C}$)
- Organic solvents and Urea
- Salts (Hofmeister Series)
- Interaction with surface (also in adsorption)

most stabilizing
strongly hydrated anions

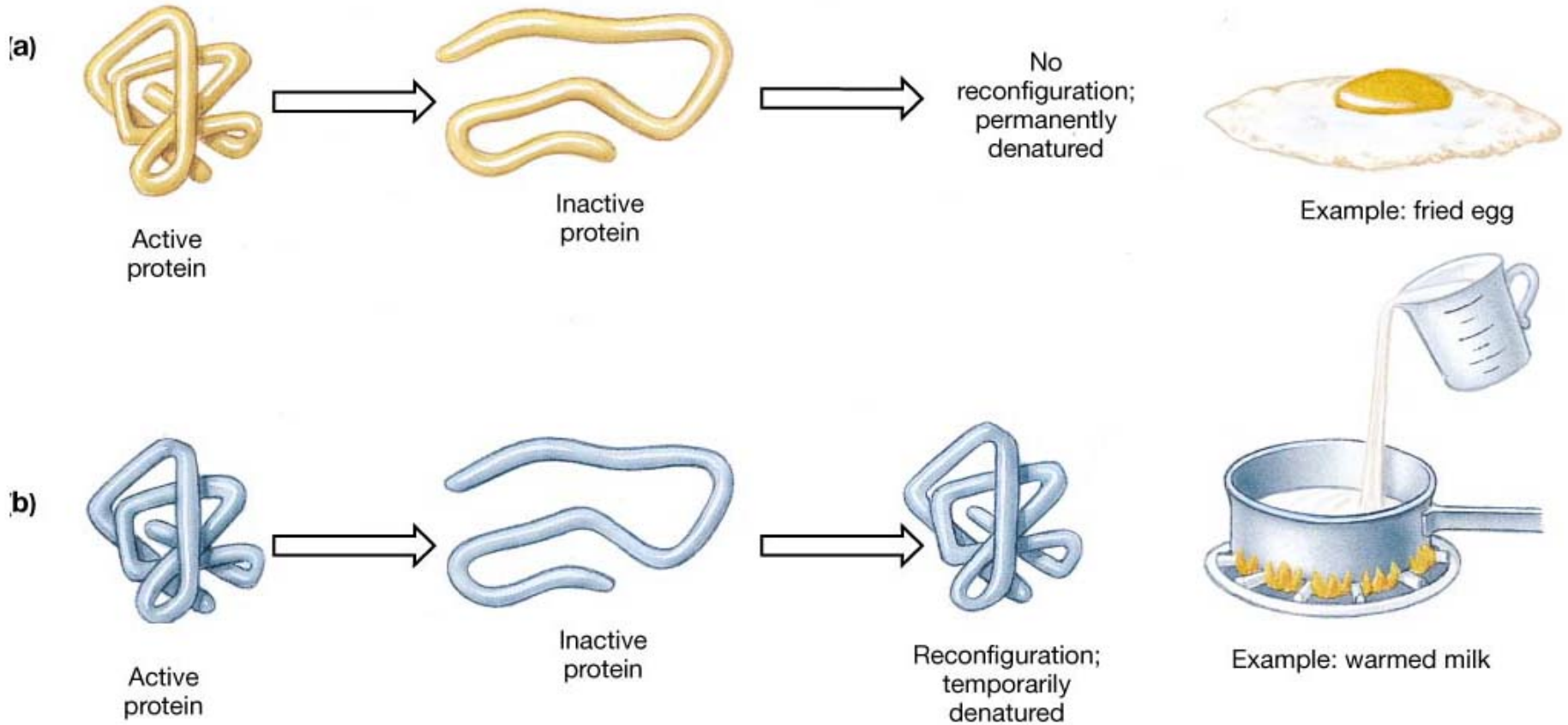
$\text{citrate}^{3-} > \text{sulfate}^{2-} > \text{phosphate}^{2-} > \text{F}^{-} > \text{Cl}^{-} > \text{Br}^{-} > \text{I}^{-} > \text{NO}_3^{-} > \text{ClO}_4^{-}$

most destabilizing
weakly hydrated anions

$\text{N}(\text{CH}_3)_4^{+} > \text{NH}_4^{+} > \text{Cs}^{+} > \text{Rb}^{+} > \text{K}^{+} > \text{Na}^{+} > \text{H}^{+} > \text{Ca}^{2+} > \text{Mg}^{2+} > \text{Al}^{3+}$

weakly hydrated cations

strongly hydrated cations



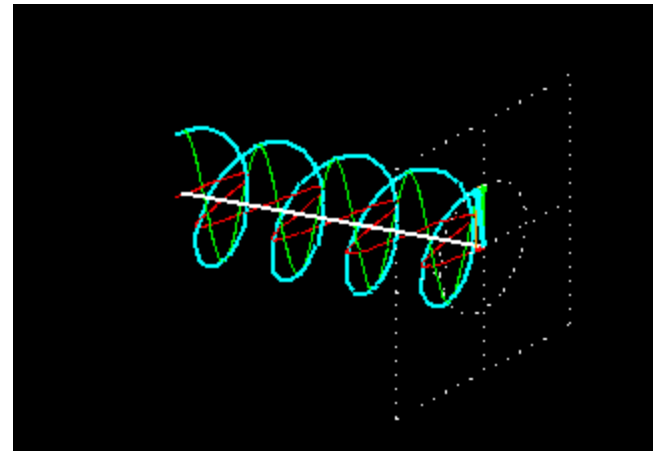
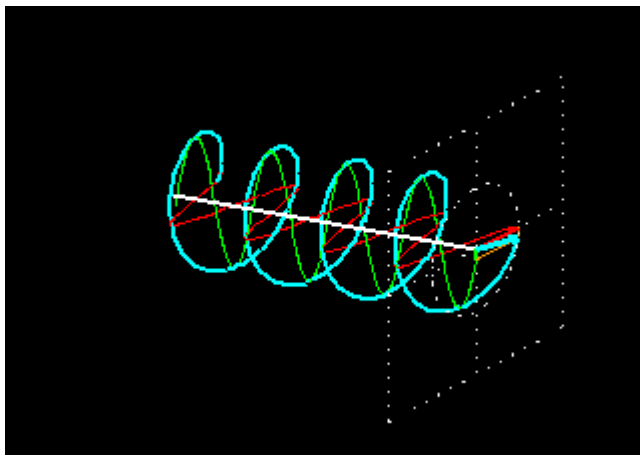
- Proteins are in equilibrium between two states: native and unfolded
- Usually reversible process
- Hydrophobic a.a. hide in the core
- Hydrophilic a.a. form a coat

How to study denaturation?

- Denaturation occurs both *in vitro* and *in vivo*
 1. Circular dichroism
 2. Fluorescence spectroscopy
 3. NMR spectroscopy

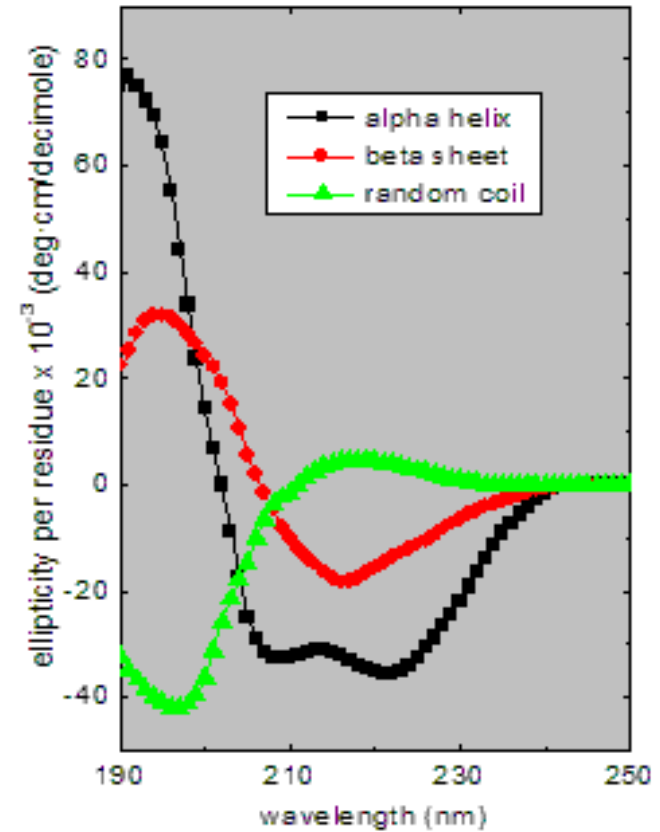
Circular dichroism (CD)

- Part of UV absorption spectroscopy
- Determines secondary structures of proteins
- Excellent to study changes in protein conformation



Circular dichroism

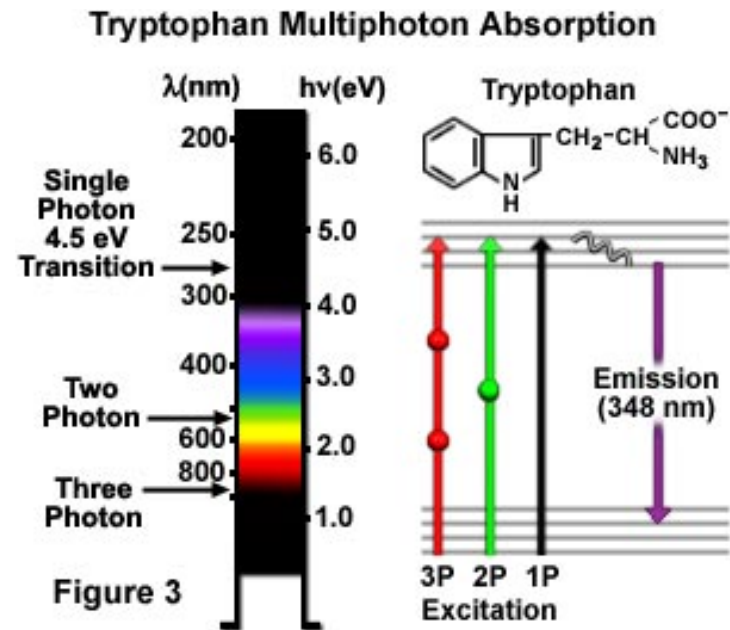
- CD spectrum reflects the structure of:
 - α -helices
 - β -sheets
 - Random coils



http://www.ap-lab.com/circular_dichroism.htm

Fluorescence spectroscopy

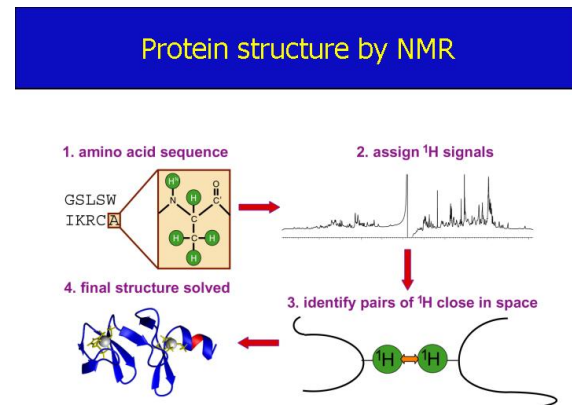
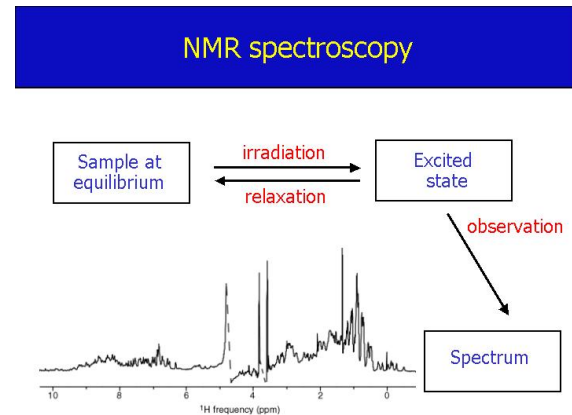
- Emissions spectroscopy
- Monitors quenching of tryptophan
- Denaturation changes the conformation of proteins leading spatial rearrangement of a.a.
- Trp exposure to electromagnetic beam is changed i.e. change in emission detected



<http://micro.magnet.fsu.edu/primer/techniques/fluorescence>

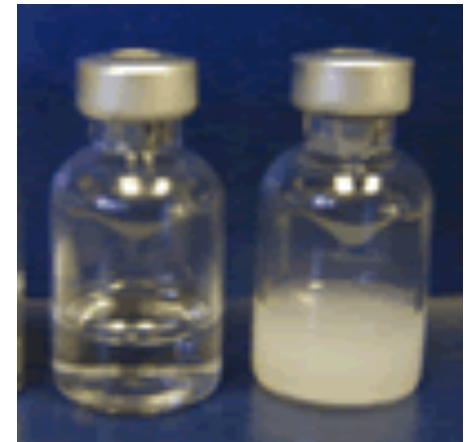
NMR spectroscopy

- Gives signals in response to proton spinning
- Each proton in a spatial conformation gives different spin i.e. different signal
- Upon exposure to magnetic field, protons spinning is changed to give different peaks
- Each peak represent a single proton which could be identified
- 2D and 3D NMR is available



Aggregation

- Partially unfolded proteins tends to accumulate
- Aggregated forms are more stable thermodynamically
- Aggregates might remain soluble or precipitate
- Adsorption enhances aggregation phenomenon
- Aggregation occurs *in vitro* and *in vivo*



Aggregation

- Physical interactions:
 - Hydrophobic interaction
- Chemical interaction:
 - Disulfide bond formation
- Mechanical agitation
 - Strong shaking



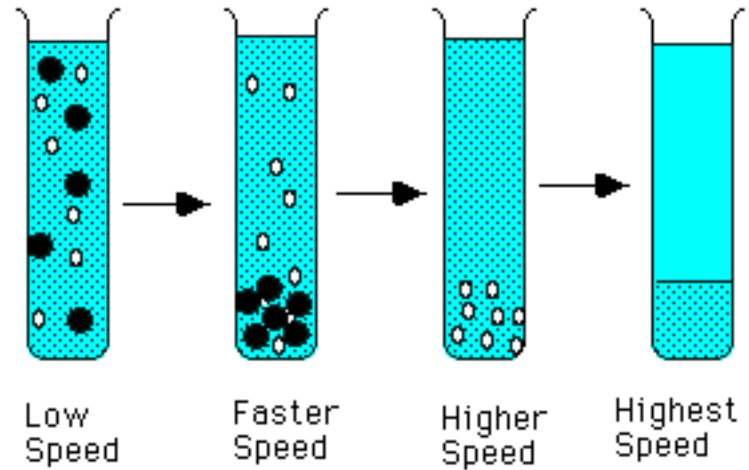
<http://www.vetmed.wsu.edu>

Aggregation

Centrifugation

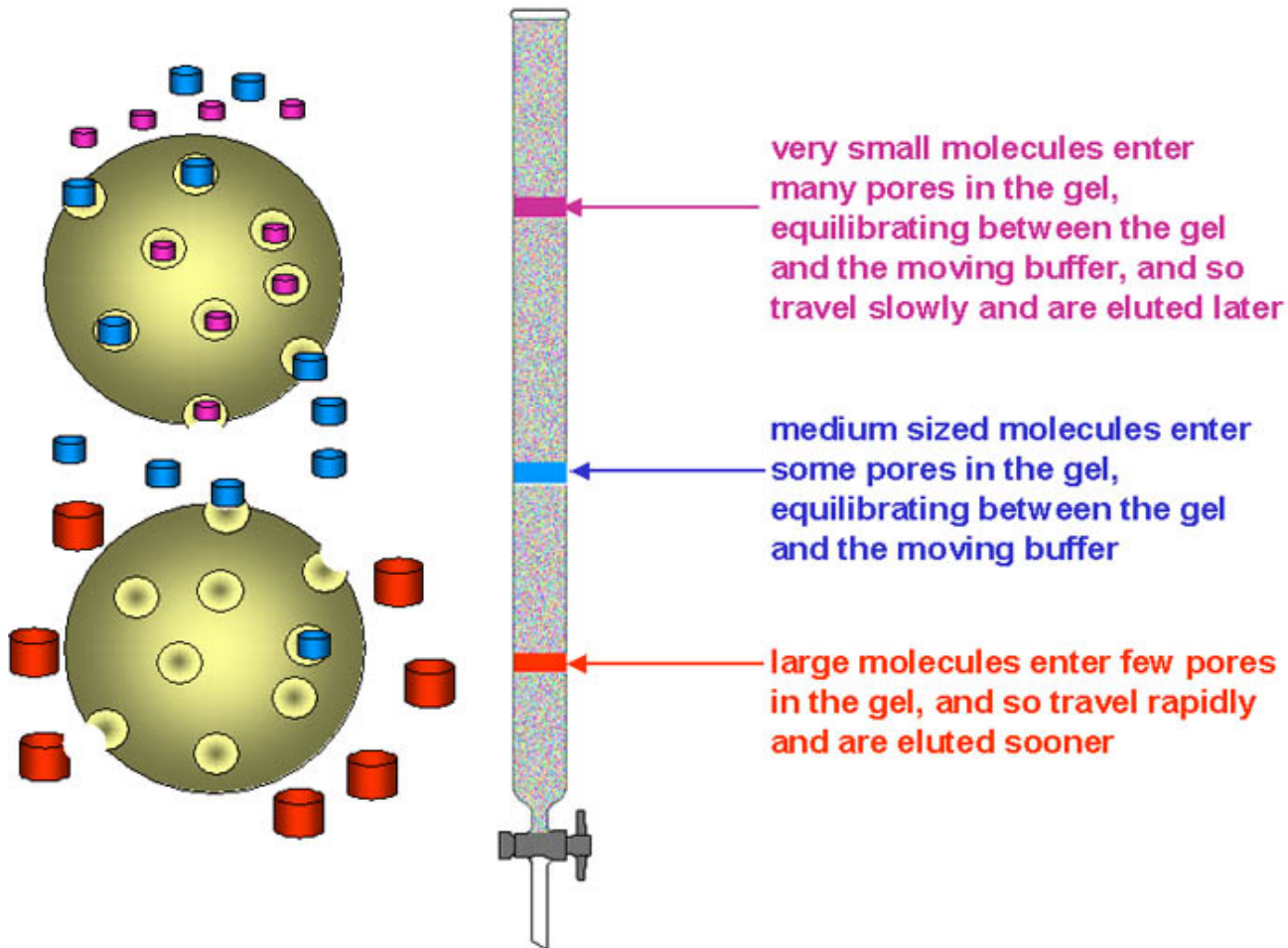


Figure 2: Differential Centrifugation.

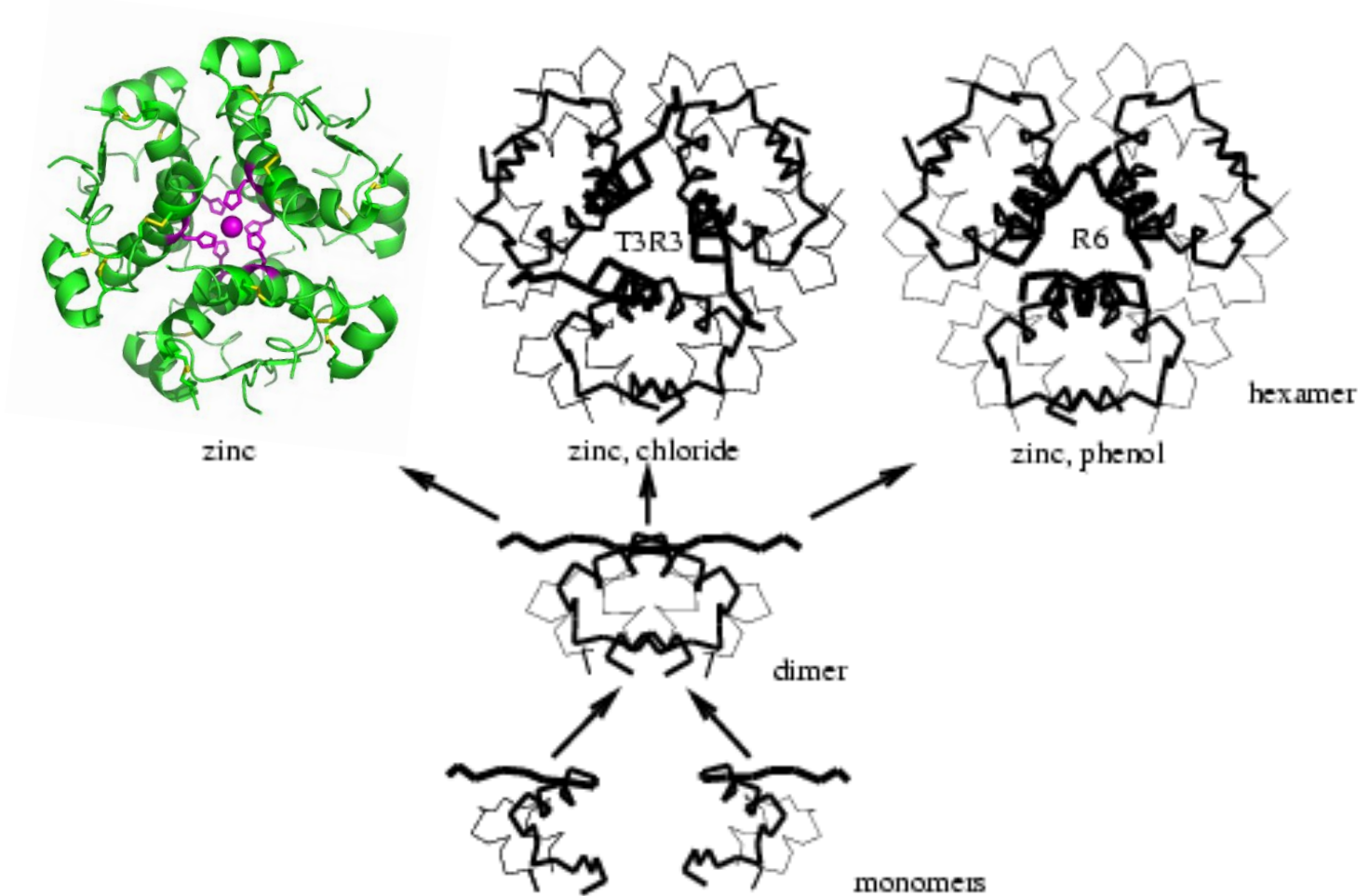


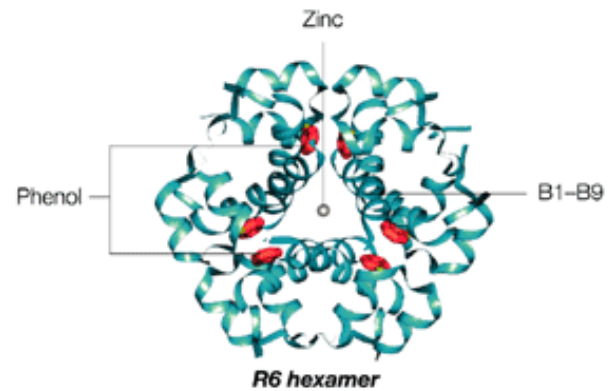
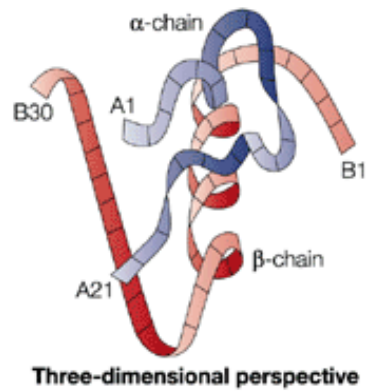
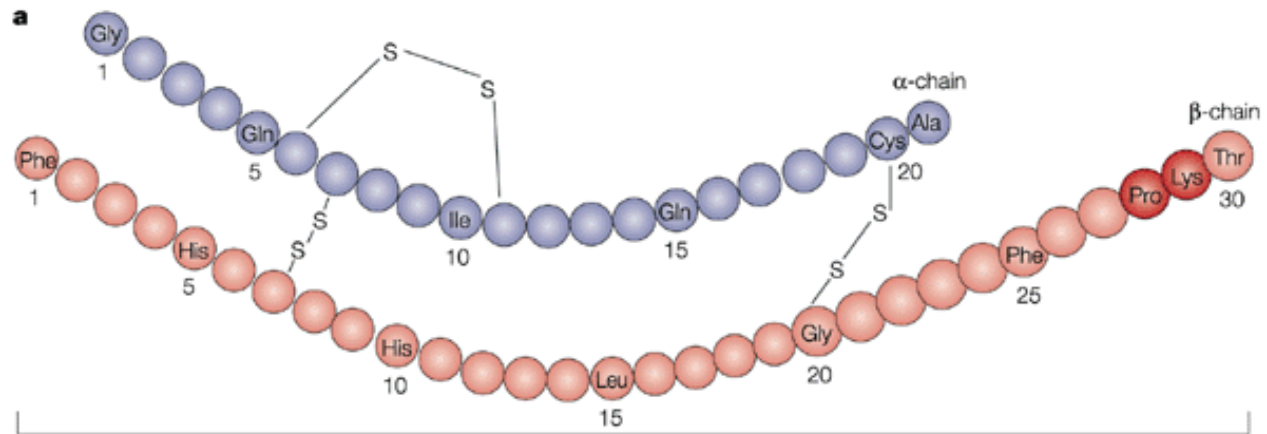
Aggregation

Size Exclusion Chromatography

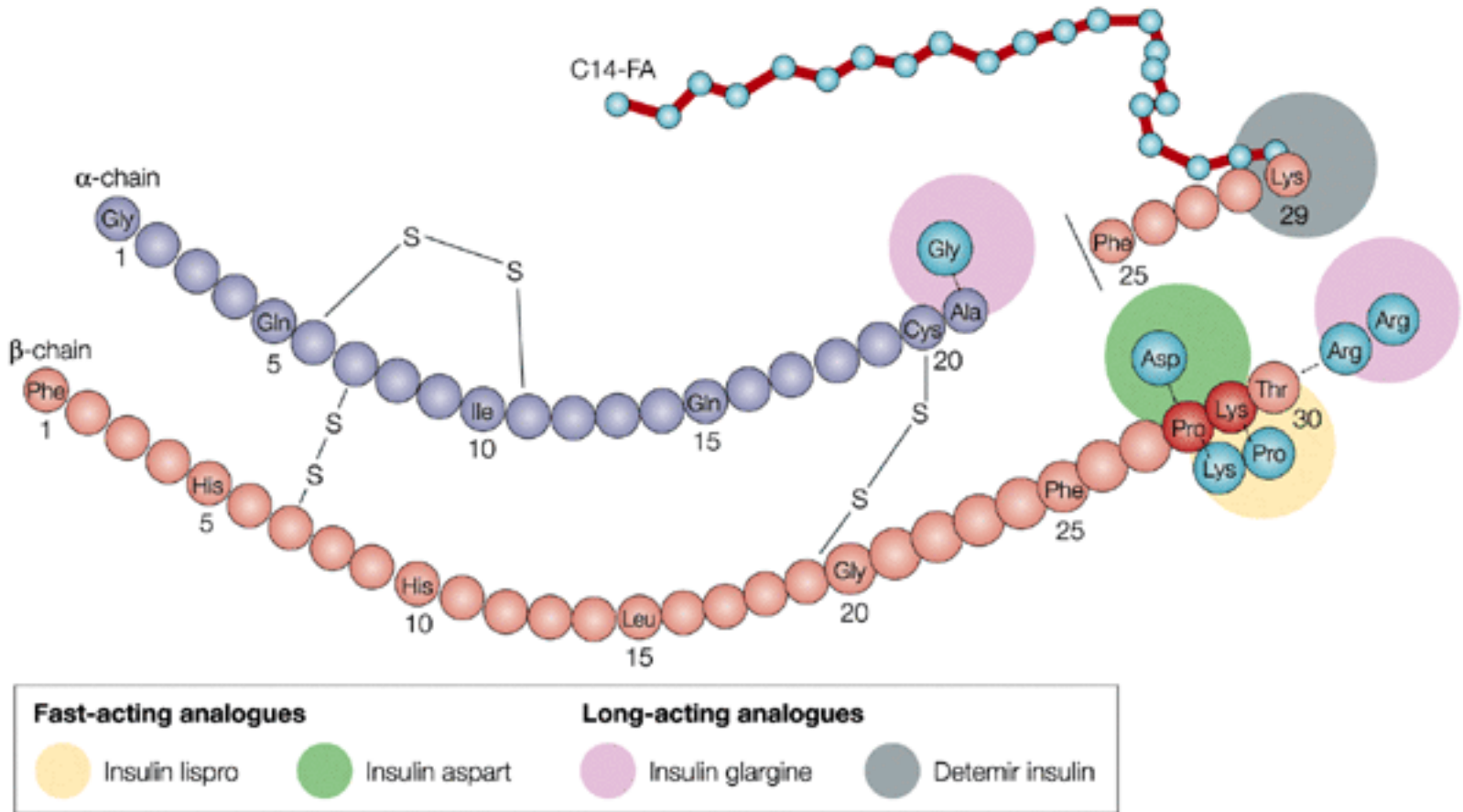


Insulin Aggregates

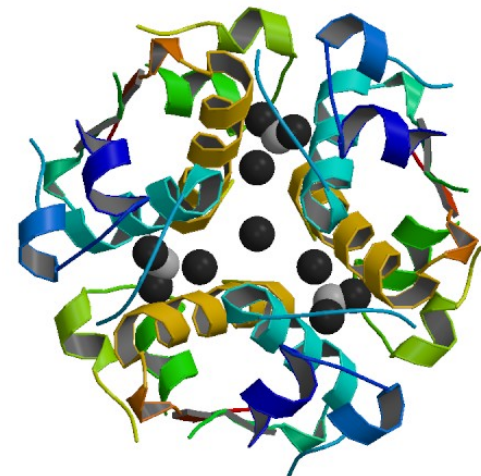
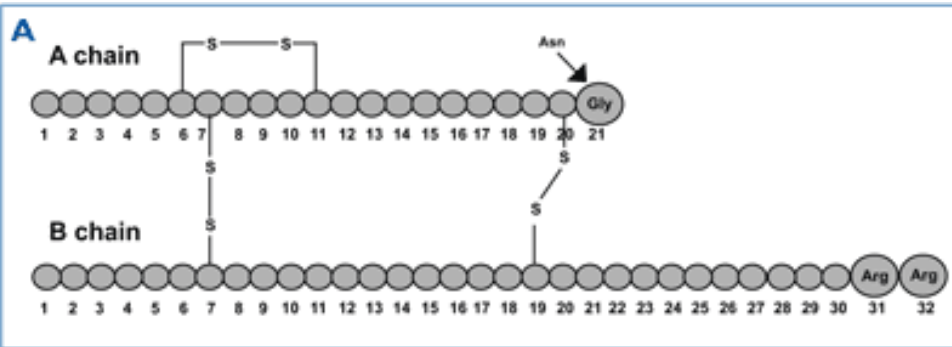
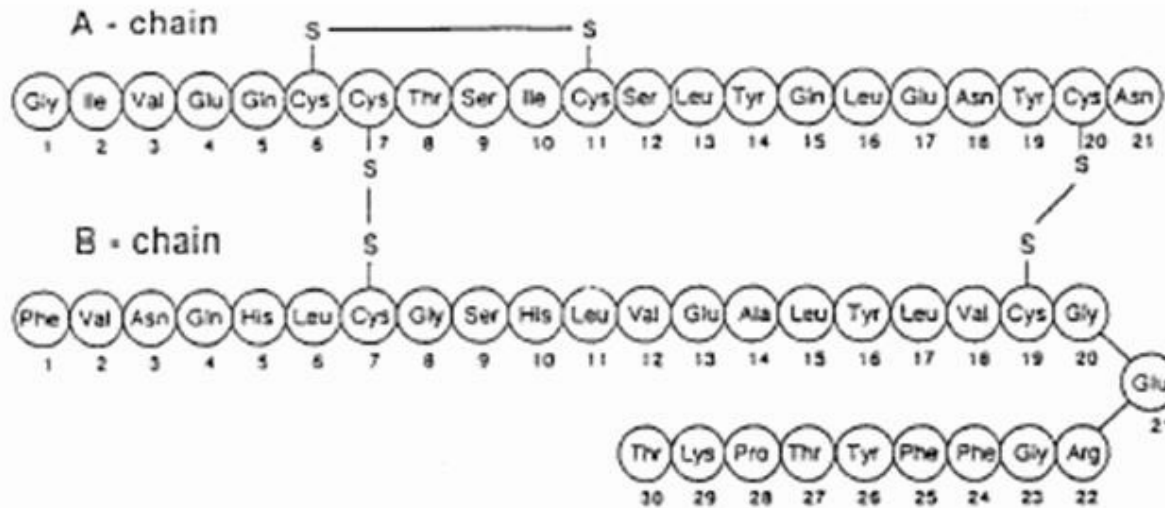




Types of Insulin



Insulin Glargine



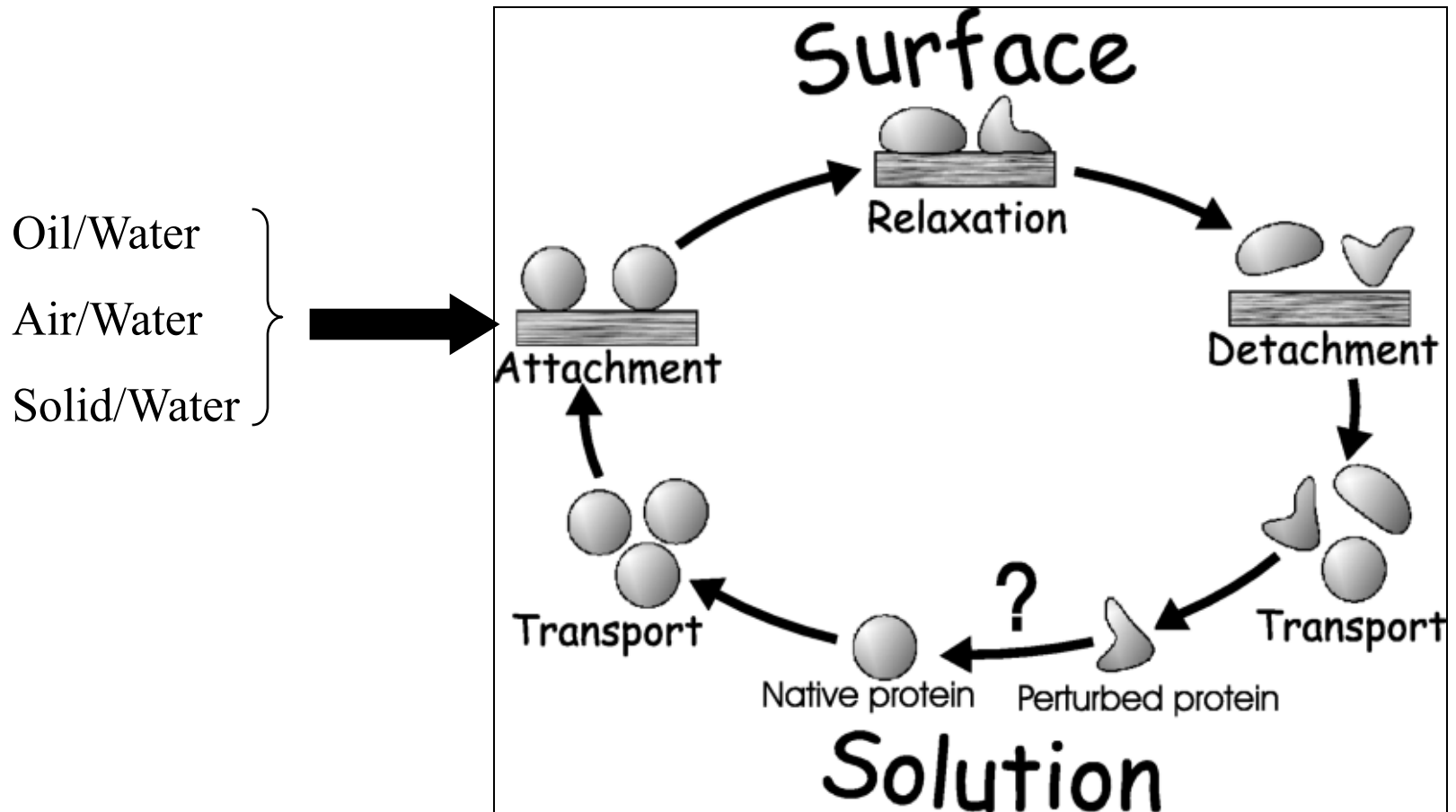
Insulin Aggregation

- The tendency of insulin to form aggregates causes problems in the diabetic patient, because of the slow release of insulin from the injection site.
- Genetically engineered insulins are therefore being produced which are not aggregated after injection and therefore quickly released into the blood stream.

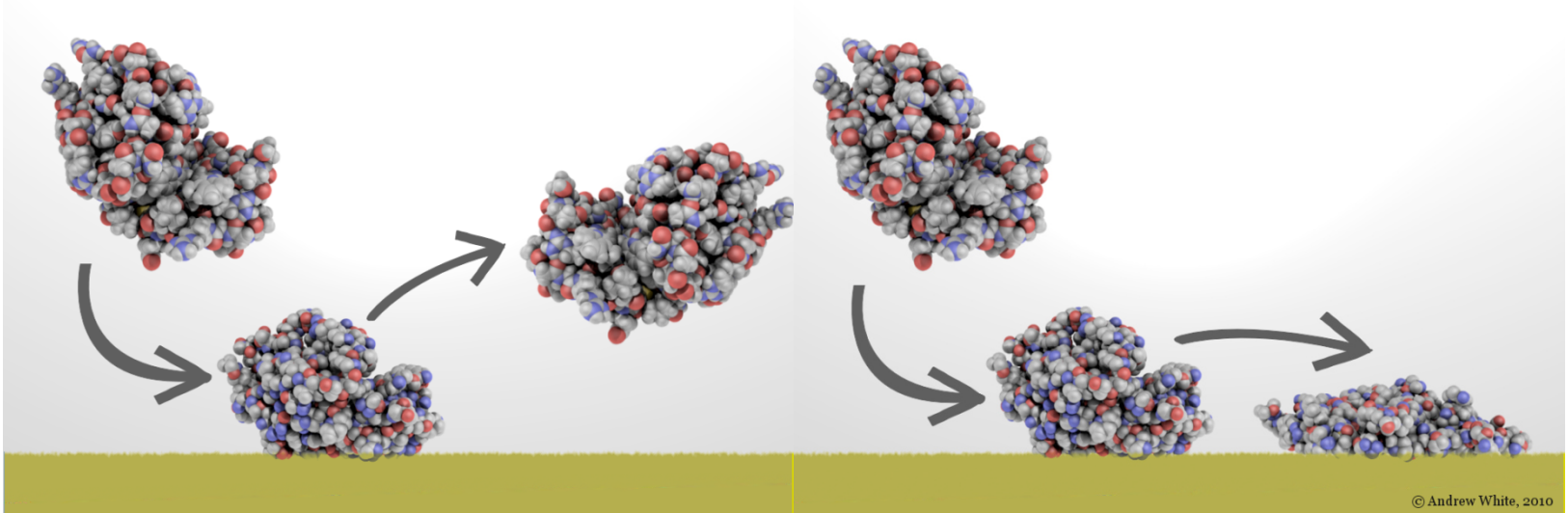
Precipitation

- Consequence of aggregation
- Large aggregates are no more soluble
- Might occur along with denaturation
- Adsorption could induce precipitation
- Stabilizing salts, pH, protein concentration, zinc ...
- Detected by “Turbidity Measurement” i.e. absorbance at 600 nm

Adsorption



Adsorption



- **Hydrophobic**

- **Electrostatic**

Now you are able to:

- ✓ Describe the challenges in pharmaceutical proteins production
- ✓ Distinguish between the different mechanisms of protein instability
- ✓ Predict the mechanism of degradation from peptide primary structure