

CLS 431 - Lecture 2

Classification of Enzymes

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Chemical Reactions - Types

 A chemical reaction is a process that is usually characterized by a chemical change in which the starting materials (reactants) are different from the products.

 Chemical reactions tend to involve the motion of electrons, leading to the formation and breaking of chemical bonds.

 There are several different types of chemical reactions and more than one way of classifying them.

Chemical Reactions - Types

- The entire chemistry of life can be "shortlisted" into 6 different types of chemical reactions:
- Redox reactions (transfer of electrons)
- 2. Transfer of functional groups
- 3. Hydrolysis
- 4. Chemical decomposition

cleavage of covalent bonds

- 5. Isomerization: structural rearrangements (without changing the net atomic composition)
- 6. Joining molecules together (ligation); combination

Enzyme classification

- The International Union of Biochemistry and Molecular Biology have developed a nomenclature for enzymes, the EC numbers
- Each enzyme is described by a sequence of four numbers preceded by "EC". The first number broadly classifies the enzyme based on its mechanism.

Major enzyme classes

- **EC 1** *Oxidoreductases*: catalyze oxidation/reduction reactions
- **EC 2** *Transferases*: transfer a functional group (*e.g.* a methyl or phosphate group)
- **EC 3** *Hydrolases*: catalyze the hydrolysis of various bonds
- EC 4 Lyases: cleave various bonds by means other than hydrolysis and oxidation
- **EC 5** *Isomerases*: catalyze isomerization changes within a single molecule
- EC 6 Ligases: join two molecules with covalent bonds.

Group	Reaction catalyzed	Typical reaction	Example(s)
EC 1 Oxidoreductases	To catalyze oxidation/reduction reactions; transfer of H and O atoms or electrons from one substance to another	 AH + B → A + BH (reduced) A + O → AO (oxidized) 	dehydrogenase, oxidase
EC 2 Transferases	Transfer of a functional group from one substance to another. The group may be methyl-, acyl-, amino- or phosphate group	$AB + C \rightarrow A + BC$	transaminase, kinase
EC 3 Hydrolases	Formation of two products from a substrate by hydrolysis	$AB + H_2O \rightarrow AOH + BH$	lipase, amylase, peptidase
EC 4 Lyases	Non-hydrolytic addition or removal of groups from substrates. C-C, C-N, C-O or C-S bonds may be cleaved	- RCOCOOH → RCOH + CO ₂ - [X-A-B-Y] → [A=B + X-Y]	decarboxylase
EC 5 Isomerases	Intramolecule rearrangement, (i.e.) isomerization changes within a single molecule	AB → BA	isomerase, mutase
EC 6 Ligases	Join together two molecules by synthesis of new C-O, C-S, C-N or C-C bonds with simultaneous breakdown of ATP	$X + Y + ATP \rightarrow XY + ADP + Pi$	synthetase

Enzyme Classes

TABLE 10.1 • Summary of the Enzyme Classes and Major Subclasses

1. Oxidoreductases

Dehydrogenases

Oxidases ·

Reductases will be student and

Peroxidases

Catalase

Oxygenases

Hydroxylases

3. Hydrolases

Esterases

Glycosidases

Peptidases

Phosphatases

Thiolases

Phospholipases

Amidases

Deaminases

Ribonucleases

5. Isomerases

Racemases

Epimerases

Isomerases

Mutases (not all)

2. Transferases

Transaldolase

and transketolase

Acyl, methyl,

glucosyl, and

phosphoryltransferases

Kinases

Phosphomutases

4. Lyases harrican amanagement

Decarboxylases

Aldolases

Hydratases

Dehydratases

Synthases

Lyases

6. Ligases

Synthetases

Carboxylases