# QUANTUM MECHANICS: LECTURE 5

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### Abstract

This lecture discusses the formal solution of Schrödinger's equation for a free particle. Including the separation of variables in stationary states, the various representations of the state vector.

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## POSITION REPRESENTATION AND THE WAVEFUNCTION

We start naturally from the postulates of quantum mechanics, having the state ket  $|\psi(t)\rangle$  ( in Schrödinger's picture) that evolves in time by Schrödinger's equation:

$$\hat{H}|\psi(t)\rangle = i\hbar \frac{\partial}{\partial t}|\psi(t)\rangle$$
 (1)

What we are interested in knowing for a free particle is its position, we therefore project the state ket into the position space , and get the **wavefunction**:

$$\psi(x,t) = \langle x | \psi(t) \rangle \tag{2}$$

The Hilbert space is therefore :  $\mathcal{H} \stackrel{\text{def}}{=} L^2(\ ] - \infty, +\infty[; dx])$  The position operator is simply a multiplicative operator :

$$\hat{X}\psi(x,t) = x\psi(x,t) \tag{3}$$

It has a continuous spectrum of eigenvalues being the position(s) of the quantum particle in the 1-dimensional space. Generalisation to 3-D space is straightforward - see homework- . The task now is to find the momentum operator in the position representation, this can be done from investigating the canonical commutation relation  $[\hat{X}, \hat{p}_x] = i\hbar \hat{I}$ , as following:

$$[\hat{X}, \hat{p}_x]\psi(x,t) = i\hbar\psi(x,t)$$
  
$$x\hat{p}_x\psi(x,t) - \hat{p}_x(x\psi(x,t)) = i\hbar\psi(x,t)$$
(4)

Rearranging the above equation :

$$\hat{p}_{x}(x\psi(x,t)) = -i\hbar\psi(x,t) + x\hat{p}_{x}\psi(x,t)$$

$$\Leftrightarrow \quad \hat{p}_{x}(x\psi(x,t)) = \frac{\hbar}{i}\frac{\partial}{\partial x}(x\psi(x,t))$$
(5)

Hence we conclude that the momentum operator in the postion representation is given by :

$$\hat{p}_x \stackrel{\text{def}}{=} \frac{\hbar}{i} \frac{\partial}{\partial x}$$
(6)

This Hilbert space implies that the particle could be anywhere in the universe until measured; but the propability vanishes at infinity points Now, we attempt to quantise the free particle system. This is done first by defining the wavefunction and the Hilbert space. Now what is left is to quantise the Hamiltonian; recall that the Hamiltonian for the free particle is given by:

$$H(p) = \frac{p^2}{2m} \tag{7}$$

The Hamiltonian operator that acts on the free-particle Hilbert space is  $\hat{H} = \frac{\hat{p}^2}{2m}$ , since we have the position representation for the momentum operator  $\hat{p}$ . We have the Hamiltonian operator:

$$\hat{H} \stackrel{\text{def}}{=} -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \tag{8}$$

Plunging it in the Schrödinger's equation; to obtain:

$$-\frac{\hbar^2}{2m}\frac{\partial^2\psi(x,t)}{\partial x^2} = i\hbar\frac{\partial\psi(x,t)}{\partial t}$$
(9)

This equation is familiar from the previous modern physics course. Nevertheless, this time we have derived it from axiomatic point of view.

In order to solve (9) we need to use a mathematical trick known as the **separation of variables**. Assume that we can write the wavefunction as the product of two functions:

$$\psi(x,t) = \varphi(x)h(t) \tag{10}$$

Substituting in (9), and rearranging, we obtain:

$$-\frac{\hbar^2}{2m}\frac{1}{\varphi(x)}\frac{d^2\varphi(x)}{dx^2} = i\hbar\frac{1}{h(t)}\frac{dh(t)}{dt}$$
(11)

Each side of the equation (11) depends only on one variable, and since they equal each other. This implies :

$$-\frac{\hbar^2}{2m}\frac{1}{\varphi(x)}\frac{d^2\varphi(x)}{dx^2} = \text{Const.}$$
(12a)

$$i\hbar \frac{1}{h(t)} \frac{dh(t)}{dt} = \text{Const.}$$
 (12b)

Evidently, the 'constant' is indeed the eigen-energy of the particle *E*. We start by solving the second equation (12b), the solution yields, what-so-called the **stationary states** time evolution:

$$h(t) = e^{-i\omega t}.$$
(13)

with  $\omega = \frac{E}{\hbar}$  Observe this result can be obtained directly from the Heisenberg picture ( Show how !) All systems of which we can separate their time dependence in this way is called stationary states. They shall be the main focus in our course.

### THE FREE-PARTICLE SOLUTION

We now turn to the spacial part of Schrödinger's equation:

$$-\frac{\hbar^2}{2m}\frac{d^2\varphi(x)}{dx^2} = E\varphi(x) \tag{14}$$

This has a particular solution of the form:

$$u(x) = Ce^{ikx} \tag{15}$$

similarity between Schrödinger's equation and the classical wave equation

observe the

Recall this from the quantisation of energy in modern physics With  $k^2 = \frac{2mE}{\hbar^2}$ , having the dimension of inverse length; we recognise *k* being the wavenumber. The solution (15) can be written in terms of the momentum - by the relation  $p = \hbar j$ .-:

$$u(x) = Ce^{\frac{1}{\hbar}px} \tag{16}$$

This solution is known as the plane wave solution. It represents a wave propagating in the +*x* direction. This solution can be used to find  $\varphi(x)$  by the superposition principle; the 'constant' of integration is not constant in fact. Rather, it is a function of *p*. In order to see this, recall that the eigen-energy of the free particle  $E = \frac{p^2}{2m} = \frac{(\hbar k)^2}{2m}$  putting this in (9) we shall have a continuous spectrum of eigen-energies. Then use the spectral theorem we shall have therefore :

$$\varphi(x) = \frac{1}{\sqrt{2\pi\hbar}} \int_{-\infty}^{+\infty} \tilde{\varphi}(p) e^{\frac{i}{\hbar}px} dp \tag{17}$$

Which is simply the Fourier transform of the momentum wavefunctions  $\tilde{\varphi}(p) = \langle p | \varphi \rangle$ . We then identify the plane-wave solution as  $\langle x | p \rangle = e^{\frac{i}{\hbar}px}$ . We can use the wavefunctions to calculate the probability of finding the particle as a given position x',  $|\varphi(x')|^2$  or momentum p',  $|\tilde{\varphi}(p')|^2$ .

Since momentum cannot be determined with absolute certainty; it then takes a Gaussian wavefunction. Of which its Fourier transform is a Gaussian function itself. The figure 3 is generated by a code we have made showing the time evolution of the wavefunction  $\psi(x,t)$ , given an initial width for the momentum wavefunction Gaussian : Observe how the wavefunction -

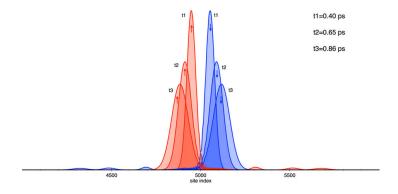


Figure 1: Simulated time evolution of a Gaussian wavefunction for the free particle, observe the dispersion of the wave as time progresses

indicating our certainty of the particle's position- gets wider and wider as time progresses. The equation (17) indicates that the position wavefunction is composed of infinite number of momentum wavefunctions. The movement of the wuantum particle is therefore expressed in terms of a **wavepacket** resulting from infinite number of waves interfering; we have run a computer code representing the wavepacket as seen in figure 3 :

## PROBABILITY FLUX AND DENSITY

Since  $|\psi(x,t)|^2$  gives the probability of finding the particle at any point in space, at a given time. It defines a positive real-valued function of space. It is known as the **pprobability density function**:

$$\rho(x,t) = \psi^*(x,t)\psi(x,t) \tag{18}$$

We also define the probability current density

$$j(x,t) = \frac{\hbar}{2mi} \left( -\frac{\partial \psi^*(x,t)}{\partial x} \psi(x,t) + \psi^*(x,t) \frac{\partial \psi(x,t)}{\partial x} \right)$$
(19)

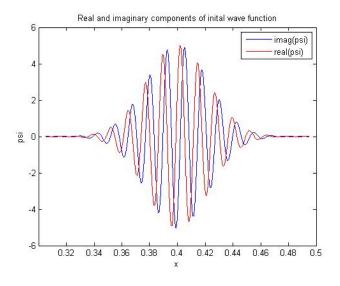


Figure 2: The wavepacket of the free quantum particle  $\psi(x, t)$ 

let's now look at  $\frac{\partial j(x,t)}{\partial x}$ :

$$\frac{\partial j(x,t)}{\partial x} = \frac{d}{dx} \left[ \frac{\hbar}{2mi} \left( -\frac{\partial \psi^*(x,t)}{\partial x} \psi(x,t) + \psi^*(x,t) \frac{\partial \psi(x,t)}{\partial x} \right) \right] \\
= \frac{\hbar}{2mi} \left( -\frac{\partial^2 \psi^*(x,t)}{\partial x^2} \psi(x,t) - \frac{\partial \psi^*(x,t)}{\partial x} \frac{\partial \psi(x,t)}{\partial x} + \psi^*(x,t) \frac{\partial^2 \psi(x,t)}{\partial x^2} + \frac{\partial \psi^*(x,t)}{\partial x} \frac{\partial \psi(x,t)}{\partial x} \right) \\
= \frac{\hbar}{2mi} \left( -\frac{\partial^2 \psi^*(x,t)}{\partial x^2} \psi(x,t) + \psi^*(x,t) \frac{\partial^2 \psi(x,t)}{\partial x^2} \right) \tag{20}$$

We can also calculate  $\dot{\rho}(x, t)$ :

$$\dot{\rho}(x,t) = \frac{\partial}{\partial t} \left( \psi^*(x,t)\psi(x,t) \right)$$
$$= \frac{\partial \psi^*(x,t)}{\partial t} \psi(x,t) + \psi^*(x,t) \frac{\partial \psi(x,t)}{\partial t}$$
(21)

But from Schrödinger's equation, we have  $\frac{\partial \psi(x,t)}{\partial t} = \frac{-\hbar}{2mi} \left( \frac{\partial^2 \psi(x,t)}{\partial x^2} \right)$ . Hence we incur:

$$\dot{\rho}(x,t) = \frac{\hbar}{2mi} \left( \frac{\partial^2 \psi^*(x,t)}{\partial x^2} \psi(x,t) - \psi^*(x,t) \frac{\partial^2 \psi(x,t)}{\partial x^2} \right)$$
(22)

$$\Leftrightarrow \frac{\partial j(x,t)}{\partial x} + \dot{\rho}(x,t) = 0$$
(23)

This is the continuity equation in quantum mechanics , implying probability is conserved .

## THE BORN CONDITIONS

Max Borns best known contribution to quantum mechanics was his proposal that the wave function, or rather its square modulus, should be interpreted as the probability density for finding the system in a given state at a given time. However, he also proposed four conditions on the wave function which are used in finding many solutions of the Schrödinger equation. As always, its useful to take another look at the Schrödinger equation (in one dimension (14)) so we can see how Born's conditions fit in.

Borns conditions to be imposed on the wave function  $\psi(x, t)$  are:

- 1. he wave function must be single valued. This means that for any given values of *x* and  $\psi(x, t)$  must have a unique value. This is a way of guaranteeing that there is only a single value for the probability of the system being in a given state.
- 2. The wave function must be square-integrable. In other words, the integral of  $|\psi|^2$  over all space must be finite. This is another way of saying that it must be possible to use  $|\psi|^2$  as a probability density, since any probability density must integrate over all space to give a value of 1, which is clearly not possible if the integral of  $|\psi|^2$  is infinite. One consequence of this proposal is that  $\psi$  must tend to o for infinite distances.
- 3. The wave function must be continuous everywhere. That is, there are no sudden jumps in the probability density when moving through space. If a function has a discontinuity such as a sharp step upwards or downwards, this can be seen as a limiting case of a very rapid change in the function. Such a rapid change would mean that the derivative of the function was very large (either a very large positive or negative number). In the limit of a step function, this would imply an infinite derivative. Since the momentum of the system is found using the momentum operator, which is a first order derivative, this would imply an infinite momentum, which is not possible in a physically realistic system. Such an infinite derivative would also violate condition 4.
- 4. All first-order derivatives of the wave function must be continuous. Following the same reasoning as in condition 3, a discontinuous first derivative would imply an infinite second derivative, and since the energy of the system is found using the second derivative, a discontinuous first derivative would imply an infinite energy, which again is not physically realistic.

For further reading on these conditions, please refer to :http://www.physicspages.com/2011/01/25/wave-function-borns-conditions/

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