



**Phys 103**

**Chapter 1**

**Physics and Measurement**

By

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# LECTURE OUTLINE

- 1.1 Standards of Length, Mass, and Time
- 1.4 Dimensional Analysis
- 1.5 Conversion of Units

# 1.1 Standards of Length, Mass, and Time

In mechanics, there are three basic quantities: length, mass, and time

- All other quantities in mechanics can be expressed in terms of these three.
- In 1960, an international committee established a set of standards for the fundamental quantities of science. It is called the SI (Système International)
- In the SI:
  - Units of length: **meter**
  - Units of mass : **kilogram**
  - Units of time : **second**

# 1.1 Standards of Length, Mass, and Time

- **Length:** SI Unit of length is: meter (m).
- **Mass:** SI Unit of mass is: kilogram (kg)
- **Time:** SI Unit of time is: second (s)
- In many situations, you may have to derive or check a specific equation. A useful and powerful procedure called dimensional analysis can be used to assist in the derivation or to check your final expression.
- As a simple method:  
Left Hand Side must = Right Hand Side

# 1.4 Dimensional Analysis

- **Dimension:** it denotes the physical nature of a quantity
- **Example:** distance: could be in meters, yards, or micrometers. But over all it is: a **length**
- **Symbols we are going to use are:**
  - dimension of length: **[L]**
  - dimension of mass: **[M]**
  - dimension of time: **[T]**

Units of Area, Volume, Velocity, Speed, and Acceleration

| System         | Area<br>(L <sup>2</sup> ) | Volume<br>(L <sup>3</sup> ) | Speed<br>(L/T) | Acceleration<br>(L/T <sup>2</sup> ) |
|----------------|---------------------------|-----------------------------|----------------|-------------------------------------|
| SI             | m <sup>2</sup>            | m <sup>3</sup>              | m/s            | m/s <sup>2</sup>                    |
| U.S. customary | ft <sup>2</sup>           | ft <sup>3</sup>             | ft/s           | ft/s <sup>2</sup>                   |

# 1.4 Dimensional Analysis

**Example:** Use dimensional analysis to check the equation:

$$x = 1/2at^2$$

• **Solution:**

$$L = \frac{L}{T^2} \cdot T^2 = L$$

**Example:** Show that  $v = at$  is dimensionally correct.

• **Solution:**

$$\text{L.H.S.:} [v] = \frac{L}{T} \quad \text{and} \quad \text{L.H.S} [at] = \frac{L}{T^2} \cdot T = \frac{L}{T}$$

$$\text{L.H.S.} = \text{R.H.S}$$

*So the equation is dimensionally correct*

# 1.5 Conversion of Units

Some times it is necessary to convert units from one measurement system to another, or to convert within a system, for example, from kilometers to meters.

**Examples:** 1 mile = 1 609 m = 1.609 km

- 1 ft = 0.304 8 m = 30.48 cm
- 1 m = 39.37 in. = 3.281 ft
- 1 in. = 0.025 4 m = 2.54 cm (exactly)

# PROBLEMS

- Section 1.4 Dimensional Analysis

**13.** The position of a particle moving under uniform acceleration is some function of time and the acceleration. Suppose we write this position

$$S = ka^m t^n$$

where  $k$  is a dimensionless constant. Show by dimensional analysis that this expression is satisfied if  $m = 1$  and  $n = 2$ . Can this analysis give the value of  $k$ ?



# PROBLEMS

- Section 1.4 Dimensional Analysis

**15.** The position of a particle moving under uniform

Which of the following equations are dimensionally correct?

*a)  $v_f = v_i + ax$*

*b)  $y = (2m) \cos(kx)$ , where  $k = 2m^{-1}$*

# PROBLEMS

- **Section 1.5 Conversion of Units**

**21.** A rectangular building lot is 100 ft by 150 ft. Determine the area of this lot in  $\text{m}^2$ .

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**25.** A solid piece of lead has a mass of 23.94 g and a volume of  $2.10 \text{ cm}^3$ . From these data, calculate the density of lead in SI units ( $\text{kg}/\text{m}^3$ ).

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**31.** One gallon of paint (volume= $3.78 \times 10^{-3} \text{ m}^3$ ) covers an area of  $25.0 \text{ m}^2$ . What is the thickness of the paint on the wall?

# Lecture Summary

- The three fundamental physical quantities of mechanics are length, mass, and time, which in the SI system have the units meters(m), kilograms(kg), and seconds(s), respectively.
- The method of dimensional analysis is very powerful in solving physics problems.
- Dimensions can be treated as algebraic quantities. By making estimates and performing order-of-magnitude calculations, you should be able to approximate the answer to a problem when there is not enough information available to completely specify an exact solution.



**Thank You**





# ACKNOWLEDGEMENTS